Name:

Jalin Shelton

Date:

3/8/23

Student Exploration: Freezing Point of Salt Water

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: freeze, freezing point, liquid, melt, melting point, solid, transformation rate

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. In the winter, people often buy large bags of rock salt to sprinkle on their walkways. Why do people do this?

it melts ice on roads and driveways

The freezing point of pure water is 0 °C (32 °F). How do you think adding salt to water affects its freezing point?

The freezing point changes.

Gizmo Warm-up

People often use salt to alter the freezing point of water. The Freezing Point of Salt Water Gizmo shows you how this works.

 With the Room temp, set to 0.0 °C, observe the water molecules in the Molecular view. Describe the motion of water molecules in the liquid phase:

The water molecules bounce off each other contiously



Set the Room temp. to -10.0 °C and observe. What is happening?

The water molecules form into solids and start vibrating. They also freeze while dropping

The process of changing from a liquid to a solid is called freezing.

Describe the motion of molecules in a solid:

They vibrate more but do move place

Freezing Point Of Saltwater Gizmo Answers

Kathleen Armour

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Freezing Point Of Saltwater Gizmo Answers Introduction

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