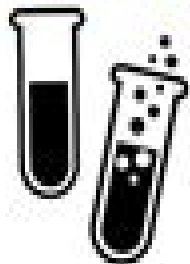


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Erdalkalimetalle	Nachmetalle
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Lanthanoide	Actinoide

					He Helium
B Boron	C Carbon	N Nitrogen	O Oxygen	F Fluorine	Ne Neon
Al Aluminium	Si Silicon	P Phosphorus	S Sulphur	Cl Chlorine	Ar Argon
Ga Gallium	Ge Germanium	As Arsenic	Se Selenium	Br Bromine	Kr Krypton
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Tl Thallium	Pb Lead	Bi Bismuth	Po Polonium	At Astatine	Rn Radon
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Erdmetalle Elemente Hauptgruppe Periodensystem Essentials

Hermann Sicius



Erdmetalle Elemente Hauptgruppe Periodensystem Essentials:

Erdmetalle: Elemente der dritten Hauptgruppe Hermann Sicius, 2015-12-05 Dieses Essential bietet eine prägnante und umfassende Übersicht über die Elemente der dritten Hauptgruppe des Periodensystems auch Borgruppe oder Erdmetalle genannt. Alle Elemente dieser Gruppe mit Ausnahme des vor etwa zehn Jahren erstmals erzeugten Ununtriums wurden erst im 19. Jahrhundert entdeckt. Obwohl Aluminium das am häufigsten vorkommende Metall ist, blieb es in elementarer Form lange verborgen. Gallium, Indium und Thallium sind dagegen sehr selten. Die Elemente dieser Gruppe ändern ihren Charakter teils stark von Halbmetall Bor über das allgegenwärtige Aluminium hin zu tiefschmelzenden Schwermetallen Gallium, Indium und Thallium und zu einem in gleicherweise leichtflüchtigen und gelegentlich anionisch auftretenden Metall Ununtrium.

Zinkgruppe: Elemente der zweiten Nebengruppe Hermann Sicius, 2017-04-24 Hermann Sicius stellt ausführlich die Elemente der zweiten Nebengruppe vor: Zink, Cadmium, Quecksilber, Copernicium, deren physikalische und chemische Eigenschaften relativ ähnlich sind. Auswirkungen der Lanthanoidenkontraktion erkennt man hier kaum. Noch Cadmium steht in seinen Eigenschaften etwa zwischen dem Zink und dem Quecksilber. Zink und Cadmium haben negative Normalpotentiale, wogegen Quecksilber ein Halbedelmetall ist. Die Elemente dieser Gruppe geben meist ein oder zwei äußere Valenzelektronen ab, um eine stabile Elektronenkonfiguration zu erreichen. Für das häufigste Element dieser Nebengruppe, das Copernicium, konnten bisher kaum chemische Untersuchungen durchgeführt werden. Es ist zu erwarten, dass es sich chemisch ähnlich wie Quecksilber verhält. Zink als Element kennt man seit dem 17. Jahrhundert, Cadmium seit 1817, wogegen Quecksilber schon in der Antike bekannt war. Die erstmalige Darstellung von Atomen des Coperniciums gelang 1996. **Chromgruppe:**

Elemente der sechsten Nebengruppe Hermann Sicius, 2016-06-23 Hermann Sicius stellt in diesem Essential die Elemente der sechsten Nebengruppe, Chromgruppe, ausführlich vor. Er zeigt, dass die Elemente dieser Gruppe chemisch meist relativ reaktionsträge sind und regelmäßige Abstufungen ihrer Eigenschaften zeigen. So nehmen vom Chrom zum Wolfram Dichte, Schmelz- und Siedepunkte zu, die Reaktivität leicht ab. Chrom, Molybdän und Wolfram entdeckte man Ende des 18. Jahrhunderts, das nur künstlich darstellbare Seaborgium erst etwa 200 Jahre später. Molybdän ist z.B. Bestandteil sehr harter Stähle, Wolfram wurde für Wendeln von Glühlampen verwendet. Vanadiumgruppe: Elemente der fünften Nebengruppe Hermann

Sicius, 2016-06-14 Hermann Sicius stellt in diesem Essential die Elemente der fünften Nebengruppe, Vanadiumgruppe, ausführlich vor. Er zeigt, dass diese Elemente teilweise reaktiv, aber gelegentlich auch sehr reaktionsträge sind, dass sie an der Luft lagernd von einer dichten passivierenden Oxidschicht vor weiterer Korrosion geschützt und auch in Säuren nur selten und dann nur unter Anwendung drastischer Methoden löslich sind. Die technische Anwendung zeigt der Autor für diese Elemente auf: die alle sehr harte, hochschmelzende Metalle sind und deren Verbindungen mit Kohlen und Stickstoff teils extrem hochschmelzende, harte Schleifmaterialien darstellen, die man auch industriell einsetzt. Vanadium findet sich in Stahlliegierungen, Tantal in medizinischen Implantaten, Ferroniob als Bestandteil von Gasturbinen oder Vanadiumdioxid als

Katalysator bei der Herstellung von Schwefelsäure **Erdalkalimetalle: Elemente der zweiten Hauptgruppe** Hermann Sicius, 2015-12-02 Hermann Sicius stellt die Elemente der zweiten Hauptgruppe des Periodensystems die Erdalkalimetalle mit ihren wichtigsten Eigenschaften und Anwendungsgebieten vor. Die Elemente dieser Gruppe Beryllium, Magnesium, Calcium, Strontium, Barium und Radium sind zwar sämtlich Metalle, zeigen aber deutliche Abstufungen ihrer Charakteristika und überraschende Eigenheiten. Beryllium ist durchlässig für Röntgenstrahlen und Barium aus derselben Gruppe nicht. Beryllium und Bariumverbindungen sind sehr giftig, Calcium und Magnesiumverbindungen für uns dagegen lebensnotwendig. In diesem Essential erfahren Sie alles Wichtige über diese interessante Hauptgruppe.

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