

7.3 Scientists traced the process of photosynthesis using isotopes

The leaves of plants that live in lakes and ponds are often covered with bubbles of O_2 produced during photosynthesis (Figure 7.3). How did scientists find out where this O_2 comes from?

The overall process of photosynthesis has been known since the 1800s. In the presence of light, green plants convert carbon dioxide and water into sugar and oxygen. Consider the basic summary equation for photosynthesis:



Looking at the equation, you can understand why scientists hypothesized that in photosynthesis, carbon dioxide is first split ($\text{CO}_2 \rightarrow \text{C} + \text{O}_2$) and O_2 released, and then water (H_2O) is added to the carbon to produce sugar. In the 1930s, this idea was challenged by C. B. van Niel, who was working with photosynthesizing bacteria that produce sugar from CO_2 but do not release O_2 in the process. These bacteria obviously did not split CO_2 in their photosynthesis. He hypothesized that in plant photosynthesis, it is H_2O that is split, with the hydrogen becoming incorporated into sugar and the O_2 released as gas.

It was almost 20 years before van Niel's hypothesis was confirmed. Using a heavy isotope of oxygen, O-18, scientists were able to follow the fate of oxygen atoms during photosynthesis. (See Module 2.3 to review isotopes.) O-18 has two more neutrons in the nucleus of its atom than the more common isotope O-16, and this slight difference in mass can be experimentally measured. To determine the source of the O_2 released by photosynthesis, researchers produced CO_2 and H_2O containing O-18, thus "labeling" the two reactant molecules. Experimental plants were provided with

one or the other labeled reactant. The results were that plants produced O_2 containing O-18 only when supplied with labeled H_2O , never when provided with labeled CO_2 . These experiments showed that the O_2 released during photosynthesis comes from water and not from CO_2 .

The synthesis of sugar in photosynthesis involves numerous chemical reactions (as you will see in Module 7.10). Working out the details of these reactions also involved the use of isotopes, in this case, radioactive isotopes. In the mid-1940s, American biochemist Melvin Calvin and his colleagues began using radioactive C-14 to trace the sequence of the intermediate molecules formed in the cyclic pathway that produces sugar from CO_2 . They worked for 10 years to elucidate this cycle, which is now called the Calvin cycle. Calvin received the Nobel Prize in 1961 for this work.



▲ Figure 7.3 Oxygen bubbles on the leaves of an aquatic plant



Photosynthesis produces billions of tons of carbohydrate a year. Where does most of the mass of this huge amount of organic matter come from?

credit: iStockphoto.com/Robert L. Taylor

7.4 Photosynthesis is a redox process, as is cellular respiration

Let's compare the processes of photosynthesis and cellular respiration. Indeed, they appear to be opposite reactions: the reactants in one are the products of the other and vice versa. Both involve oxidation-reduction reactions (see

this energy boost. Figure 7.4 illustrates these energy-requiring redox reactions as CO_2 becomes reduced to sugar and water molecules are oxidized to O_2 .

Chapter 7 Photosynthesis Study Guide

Martha Mackin



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web 1 the distance between points on the scale is equal across the range of interval data 2 ordinal data can be ordered but

interval data cannot 3 interval variables contain only two categories 4 ordinal variables have a fixed zero point whereas interval variables do not 5