CHAPTER ASSESSMENT

# Solutions

CHAPTER

#### **Reviewing Vocabulary**

Match the definition in Column A with the term in Column B.

#### Column A.

- The diffusion of solvent particles across a semipermeable membrane from areas of lower solute concentration to areas of higher solute concentration
- A mixture with particles that settle out if undisturbed
  - 3. The erratic movement of colloid particles
  - The amount of additional pressure caused by water molecules moving into a solution
  - A measure of how much solute is dissolved in a specific amount of solvent or solution
- 6. The overall energy change that occurs when a solution forms
- 7. A heterogeneous mixture of intermediate size particles
  - 8. The process of surrounding solute particles with solvent particles to form a solution
  - The ratio of the number of moles of solute in solution to the total number of moles of solute and solvent
- \_\_\_\_\_\_ 10. The scattering of light by dispersed colloid particles
  - 11. The statement that the solubility of a gas in a liquid is directly proportional to the pressure of the gas above the liquid

Describe each pair of related terms.

- 12. soluble, insoluble
- 13. miscible, immiscible
- 14. molarity, molality

Column B

- a. Brownian motion
- b. colloid
- c. concentration
- d. heat of solution
- e. Henry's law
- f. mole fraction.
- g. osmosis.
- h. osmotic pressure
- i. solvation
- j. Tyndall effect

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k, suspension

## **Chapter 15 Chemistry Study Guide Answers**

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