

Chemistry 12 Worksheet 2-3

[KEY]

Calculations Involving the Equilibrium Constant K_{eq}

1. Given the equilibrium equation below:



If, at equilibrium, the concentrations are as follows:

$$[A_2] = 3.45 \text{ M}, \quad [B_2] = 5.67 \text{ M} \quad \text{and} \quad [AB] = 0.67 \text{ M} \quad (2 \text{ SD})$$

- a) Write the expression for the equilibrium constant,
- K_{eq}

$$K_{eq} = \frac{[AB]^2}{[A_2][B_2]}$$

- b) Find the value of the equilibrium constant,
- K_{eq}
- at the temperature that the experiment was done.

$$K_{eq} = \frac{(0.67)^2}{(3.45)(5.67)} = 0.023$$

Answer $K_{eq} = 0.023$

2. Given the equilibrium equation:

at a temperature of 50°C , it is found that when equilibrium is reached that:

$$[X_2] = 0.37 \text{ M}, \quad [Y_2] = 0.53 \text{ M} \quad \text{and} \quad [XY_3] = 0.090 \text{ M} \quad (\text{All } 2 \text{ SD})$$

- a) Write the equilibrium constant expression (
- K_{eq}
-)

$$K_{eq} = \frac{[XY_3]^2}{[X_2][Y_2]^3}$$

- b) Calculate the value of
- K_{eq}
- at
- 50°C
- .

$$K_{eq} = \frac{(0.090)^2}{(0.37)(0.53)^3} = 0.15$$

Answer $K_{eq} = 0.15$

Chapter 12 Pearson Chemistry Workbook Answers

Keith Barnham



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Introduction to Geochemistry Kula C. Misra, 2012-05-21 INTRODUCTION TO Geochemistry This book is intended to serve as a text for an introductory course in geochemistry for undergraduate graduate students with at least an elementary level background in earth sciences chemistry and mathematics The text containing 83 tables and 181 figures covers a wide variety of topics ranging from atomic structure to chemical and isotopic equilibria to modern biogeochemical cycles which are divided into four interrelated parts Crystal Chemistry Chemical Reactions and biochemical reactions involving bacteria Isotope Geochemistry radiogenic and stable isotopes and The Earth Supersystem which includes discussions pertinent to the evolution of the solid Earth the atmosphere and the hydrosphere In keeping with the modern trend in the field of geochemistry the book emphasizes computational techniques by developing appropriate mathematical relations solving a variety of problems to illustrate application of the mathematical relations and leaving a set of questions at the end of each chapter to be solved by students However so as not to interrupt the flow of the text involved chemical concepts and mathematical derivations are separated in the form of boxes Supplementary materials are packaged into ten appendixes that include a standard state 298 15 K 1 bar thermodynamic data table and a listing of answers to selected chapter end questions

Fundamentals of Quantum Chemistry J. E. House, 2004 This is a self contained student friendly introduction to the key concepts of quantum chemistry The math is developed as needed and motivated by the concepts themselves Midwest

Workbook for Clinical Medical Assisting Frazier, Margaret Schell Frazier, Connie Morgan, 2008-01-05 Chemistry: The Central Science Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation

The Burning Answer Keith Barnham, 2015-05-15 Our society faces a choice We could be enjoying a sustainable lifestyle but we have chosen not to In three generations we have consumed half the oil produced by photosynthesis over eight million generations In two generations we have used half our uranium resources With threats from global warming oil depletion and nuclear disaster we are running out of options Solar power as Keith Barnham says is our necessary solution Barnham explains that the roots of solar energy lie in a little known equation $E = hf$ an equation which was coincidentally celebrated and explained to the world by Einstein in the same year he discovered $E = mc^2$ He alleges

that the former equation has been overlooked in favor of the latter much to our detriment and Barnham is here to offer us a solution We can still turn things around and solar energy is the key In this provocative inspiring passionately argued book Keith Barnham outlines actions that any one and all of us can take to make an impact now and on future generations The Burning Answer is a solar manifesto for the new climate aware generation and a must read for climate change skeptics

Sif: Chemistry 5na Tb J. G. R. Briggs, 2009 *Chemistry I | AICTE Prescribed Textbook - English* Manisha

Agrawal, 2021-11-01 Chemistry I is a compulsory paper for the first year Undergraduate course in Engineering Technology Syllabus of this book is strictly aligned as per model curriculum of AICTE and academic content is amalgamated with the concept of outcome based education Book covers seven topics Atomic and molecular structure Spectroscopic Technique and applications Inter molecular Forces and Potential Energy Surfaces Use of Free Energy in Chemical Equilibrium Periodic Properties Stereo chemistry Organic Reactions and Synthesis of Drug Molecules Each topic is written in easy and lucid manner Every chapter contains a set of exercise at the end of each unit to test student's comprehension Salient Features Content of the book aligned with the mapping of Course Outcomes Programs Outcomes and Unit Outcomes Book Provides lots of recent information interesting facts QR Code for E resources QR Code for us of ICT Projects group discussion etc Students and teacher centric subject materials included in book with balanced and chronological manner Figures tables chemical equations and comparative charts are inserted to improve clarity of the topics Short questions objective questions and long answer exercises are given for practice of students after every chapter Solved and unsolved problems including numerical examples are solved with systematic steps **The Chemical News**, 1920 The Pearson General Studies

Manual 2009, 1/e Showick Thorpe Edgar Thorpe, 2009 This latest edition of The Pearson General Studies Manual continues to provide exhaustive study material for the General Studies paper of the UPSC Civil Services Preliminary Examination This student friendly book has been completely revised thoroughly updated and carefully streamlined and is strictly exam centric In this new edition a large number of new boxes and marginalia with additional and relevant information have been added to provide cutting edge information to the aspirant Readers will find that important facts and information have been presented in the form of well structured tables and lists Descriptive Inorganic Chemistry James E. House, Kathleen A.

House, 2010-09-22 Descriptive Inorganic Chemistry Second Edition covers the synthesis reactions and properties of elements and inorganic compounds for courses in descriptive inorganic chemistry This updated version includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes and incorporates new industrial applications matched to key topics in the text It is suitable for the one semester ACS recommended course or as a supplement in general chemistry courses Ideal for majors and non majors the book incorporates rich graphs and diagrams to enhance the content and maximize learning Includes expanded coverage of chemical bonding and enhanced treatment of Buckminster Fullerenes Incorporates new industrial applications matched to key topics in the text *Chemical News and Journal of Physical Science*

,1920 *Chemical News and Journal of Industrial Science* ,1919 **Chemical News and Journal of Physical Science**
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