

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. It increases in the periodic table.
- _____ 2. It decreases in the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are shiny and conduct electricity.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 1. There are two representative elements of elements.
- _____ 2. There are two periods of the elements in the periodic table.
- _____ 3. Group 1 elements include the noble gases and elements in the alkali metals.
- _____ 4. Group 17 elements are the alkali metals.
- _____ 5. Group 1 elements are highly reactive transition elements in the periodic table.
- _____ 6. Group 17 elements are very reactive elements found in the noble gases.
- _____ 7. Alkali metals have properties of both metals and non-metals.

Chapter 6 Study Guide For Content Mastery

Kathleen Armour



Chapter 6 Study Guide For Content Mastery:

This book delves into Chapter 6 Study Guide For Content Mastery. Chapter 6 Study Guide For Content Mastery is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chapter 6 Study Guide For Content Mastery, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 1: Introduction to Chapter 6 Study Guide For Content Mastery
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