

12

STOICHIOMETRY

Practice Problems

In your notebook, solve the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

1. An apple pie needs 10 large apples, 2 cups (top and bottom), and 1 tablespoon of cinnamon. Write a balanced equation that fits this situation. How many apples are needed to make 20 pies?
2. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate, $\text{KClO}_3(\text{s})$. Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate?
4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from 3.8×10^{23} molecules of oxygen? How many moles of water are produced from 22.5 moles of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

1. Calculate the number of moles of hydrogen chloride produced from 10 moles of hydrogen.



2. Calculate the number of moles of chlorine needed to form 14 moles of iron(II) chloride.



3. Calculate the number of grams of nitrogen dioxide that are produced from 4 moles of nitric oxide.



4. Calculate the mass of oxygen produced from the decomposition of 75.0 g of potassium chlorate.



5. Calculate the mass of silver needed to react with chlorine to produce 94 g of silver chloride. Write a balanced equation first.

6. How many liters of carbon monoxide at STP are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



7. Calculate the number of liters of oxygen gas needed to produce 15.0 liters of dinitrogen monoxide. Assume all gases are at the same conditions of temperature and pressure.



Chapter 12 Stoichiometry Guided Reading Answer Key

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. Teaching Science for Understanding James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities. **Resources in Education**, 1980-04

CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end. Two full length practice tests with detailed answer explanations are included in the book.

Books in Print Supplement, 2002 *Resources in Education*, 1980 Holt Chemistry Ralph Thomas Myers, 2004

Children's Books in Print R R Bowker Publishing, Bowker, 1999-12 **Children's Books in Print, 2007**, 2006

Cambridge IGCSE™ Chemistry Student's Book (Collins Cambridge IGCSE™) Chris Sunley, 2021-04-22 Collins IGCSE Chemistry provides complete coverage of the latest Cambridge IGCSE syllabus for Chemistry and is packed full of questions in depth, content, practical, investigative skills, features, and more. CAS Curriculum Advisory Service, Inc, 1974

Biochemistry Explained Thomas Millar, 2018-12-14 Biochemistry Explained employs an innovative approach which has proven highly successful in the author's own classes. The author establishes a thorough understanding of the foundations of and common linkages between molecular structures and reactions so that eventual interpretation of complex biochemical pathways and reactions is easy. All of the major molecular structures and biochemical pathways are explained, and for the most part these center on mammalian biochemistry. The text is supported by biochemical nomenclature and questions to bear in mind while reading. Higher learning sections are also provided for advanced students. Written in an informal conversational

style this textbook will serve as an invaluable resource for any student who is struggling with the standard texts and for postgraduate students who need to refresh their knowledge **CAS. Curriculum Advisory Service Quarterly** ,1973

Introduction to Experimental Biophysics Jay L. Nadeau,2017-10-10 Praise for the First Edition essential reading for any physical scientist who is interested in performing biological research Contemporary Physics an ambitious text Each chapter contains protocols and the conceptual reasoning behind them which is often useful to physicists performing biological experiments for the first time Physics Today This fully updated and expanded text is the best starting point for any student or researcher in the physical sciences to gain firm grounding in the techniques employed in molecular biophysics and quantitative biology It includes brand new chapters on gene expression techniques advanced techniques in biological light microscopy super resolution two photon and fluorescence lifetime imaging holography and gold nanoparticles used in medicine The author shares invaluable practical tips and insider s knowledge to simplify potentially confusing techniques The reader is guided through easy to follow examples carried out from start to finish with practical tips and insider s knowledge The emphasis is on building comfort with getting hands wet with basic methods and finally understanding when and how to apply or adapt them to address different questions Jay L Nadeau is a scientific researcher and head of the Biomedical Engineering in Advanced Applications of Quantum Oscillatory and Nanotechnological Systems BEAQONS lab at Caltech and was previously associate professor of biomedical engineering and physics at McGill University The Cumulative Book Index ,1909 A world list of books in the English language *Programmed Learning and Individually Paced Instruction* Carl H. Hendershot,1973 **Choice** Richard K. Gardner,Phyllis Grumm,1976 **Choice** ,2006 The United States Catalog ,1909 **The Software Encyclopedia 2001** ,2001

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