

## IB Chemistry – SL

### Topic 2 Questions

1. Consider the composition of the species W, X, Y and Z below. Which species is an anion?

Species	Number of protons	Number of neutrons	Number of electrons
W	9	10	10
X	11	12	11
Y	12	12	12
Z	13	14	10

- A. W
- B. X
- C. Y
- D. Z

(Total 1 mark)

2. Energy levels for an electron in a hydrogen atom are

- A. evenly spaced.
- B. farther apart near the nucleus.
- C. closer together near the nucleus.
- D. arranged randomly.

(Total 1 mark)

3. Which is related to the number of electrons in the outer most energy level of the elements from the alkali metals to the halogens?

- I. Group number
  - II. Period number
- A. I only
  - B. II only
  - C. Both I and II
  - D. Neither I nor II

(Total 1 mark)

4. How do bond length and bond strength change as the number of bonds between two atoms

# Chapter 14 Study Guide Mixtures Solutions

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B. Bailey**

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