



Name:

Date:

Student Exploration: Boyle's Law and Charles's Law

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: absolute zero, Boyle's law, Charles's law, Gay-Lussac's law, Kelvin scale, pressure

Prior Knowledge Question (Do this BEE OFE using the Gizmo.)

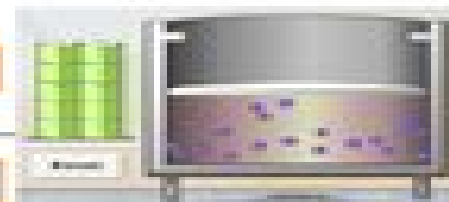
A small helium tank measures about two feet (60 cm) high. Yet it can fill over 50 balloons! How can such a small tank contain enough helium to fill so many balloons?

Gizmo Warm-up

The Boyle's Law and Charles's Law Gizmo shows a container of gas. Inside, small purple spheres represent gas molecules.

1. Observe the particles: Are they all moving at the same speed?

2. How do the particles interact with the walls and lid of the container?



These interactions contribute to the **pressure** on the walls of the container. Pressure is defined as force per unit area. The SI units of pressure are newtons per square meter (N/m^2), or pascals (Pa).

3. Slowly drag the temperature (**T**) slider back and forth. (Note: In this Gizmo, the **Kelvin scale** is used to measure temperature. On the Kelvin scale, 0 degrees is **absolute zero**, the coldest possible temperature. Absolute zero is equal to -273.15°C or -459.67°F .)

A. How does the change in temperature affect the speed of the molecules?

B. How does the change in temperature affect the volume of the container?

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N. Sue Barnes



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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