#### : Knowt

### chapter 12 study guide

1. What is the difference between anabolism, catabolism and metabolism?

anabolism: <u>synthesis of complex molecules</u> from smaller, simpler ones → <u>requires energy</u> input as these reactions are energetically unfavourable → storing energy resources in energy-rich bonds of complex molecules usually through condensation metabolism: the <u>sum and interaction of all bio-chemical processes</u> that occur within a living organism in order to <u>maintain homeostasis</u>, thus maintain life → catabolism and anabolism in interplay catabolism: <u>break down of complex molecules</u> into smaller parts, <u>energy becomes available</u> as rich rich bonds are cleaved (mostly through using water → hydrolysis) thus energetically favourable → destructive metabolism

- Be able to describe each of the four "stages" of the complete oxidation of glucose, if this occurs under aerobic or anaerobic conditions and where in the cell each occurs. Understand how the process of transferring electrons form glucose to O<sub>2</sub> results in ATP production.
  - glycolysis cytoplasm, anaerobic (1x glucose 6C → 2x pyruvate 3C)
  - PDH (pyruvate decarboxylation) mitochondrial matrix, aerobic (2x pyruvate 3C → 2x acetyl CoA 2C; production of 1 CO2 per acetyl CoA + NADH)
  - krebs cycle mt matrix, aerobic (acetyl-CoA 2C + oxaloacetate 4C → citrate 6C → production of 2 CO2, 1 ATP, 3x NADH, 2x FADH2 per round/per pyruvate)
  - oxidative phosphorylation (ETC and ATP-synthase) inner mitochondrial membrane, aerobic → NADH and FADH2 donate electrons to the ETC to generate proton motive force → towards mitochondrial matrix (through pumping

## **Chapter 12 Study Guide**

Richard O. Straub, Kathleen Stassen Berger

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Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us Organic Chemistry, Student Study Guide and Solutions Manual David R. 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