

# PERIODIC TABLE OF THE ELEMENTS

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PERIOD

PERIODIC TABLE OF THE ELEMENTS

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1 1.0079 <b>H</b> HYDROGEN	2 4.0026 <b>He</b> HELIUM																	18 4.0026 <b>He</b> HELIUM																
3 6.941 <b>Li</b> LITHIUM	4 9.0122 <b>Be</b> BERYLLIUM																	19 18.998 <b>F</b> FLUORINE	20 19.998 <b>Ne</b> NEON															
11 22.990 <b>Na</b> SODIUM	12 24.305 <b>Mg</b> MAGNESIUM	13 10.811 <b>B</b> BORON	14 12.011 <b>C</b> CARBON	15 14.007 <b>N</b> NITROGEN	16 15.999 <b>O</b> OXYGEN	17 35.453 <b>Cl</b> CHLORINE	18 36.966 <b>Ar</b> ARGON												35 79.904 <b>Br</b> BROMINE	36 83.798 <b>Kr</b> KRYPTON														
19 39.098 <b>K</b> POTASSIUM	20 40.078 <b>Ca</b> CALCIUM	21 44.956 <b>Sc</b> SCANDIUM	22 47.867 <b>Ti</b> TITANIUM	23 50.942 <b>V</b> VANADIUM	24 51.996 <b>Cr</b> CHROMIUM	25 54.938 <b>Mn</b> MANGANESE	26 55.945 <b>Fe</b> IRON	27 58.933 <b>Co</b> COBALT	28 58.933 <b>Ni</b> NICKEL	29 63.546 <b>Cu</b> COPPER	30 65.38 <b>Zn</b> ZINC	31 69.723 <b>Ga</b> GALLIUM	32 72.64 <b>Ge</b> GERMANIUM	33 74.922 <b>As</b> ARSENIC	34 78.96 <b>Se</b> SELENIUM	35 79.904 <b>Br</b> BROMINE	36 83.798 <b>Kr</b> KRYPTON																53 126.90 <b>I</b> IODINE	54 131.29 <b>Xe</b> XENON
37 85.468 <b>Rb</b> RUBIDIUM	38 87.62 <b>Sr</b> STRONTIUM	39 88.906 <b>Y</b> YTIUM	40 91.224 <b>Zr</b> ZIRCONIUM	41 92.906 <b>Nb</b> NIOBIUM	42 95.96 <b>Mo</b> MOLYBDENUM	43 98 <b>Tc</b> TECHNETIUM	44 101.07 <b>Ru</b> RUTHENIUM	45 102.91 <b>Rh</b> RHODIUM	46 106.42 <b>Pd</b> PALLADIUM	47 107.87 <b>Ag</b> SILVER	48 112.41 <b>Cd</b> CADMIUM	49 114.82 <b>In</b> INDIUM	50 118.71 <b>Sn</b> TIN	51 121.76 <b>Sb</b> ANTIMONY	52 127.60 <b>Te</b> TELLURIUM	53 126.90 <b>I</b> IODINE	54 131.29 <b>Xe</b> XENON																85 210 <b>At</b> ASTATINE	86 222 <b>Rn</b> RADON
55 132.91 <b>Cs</b> CAESIUM	56 137.33 <b>Ba</b> BARIUM	57-71 La-Lu LANTHANIDE	72 178.49 <b>Hf</b> HAFNIUM	73 180.90 <b>Ta</b> TANTALUM	74 183.84 <b>W</b> TUNGSTEN	75 186.21 <b>Re</b> RHENIUM	76 186.23 <b>Os</b> OSMIUM	77 190.23 <b>Ir</b> IRIDIUM	78 195.08 <b>Pt</b> PLATINUM	79 196.97 <b>Au</b> GOLD	80 200.59 <b>Hg</b> MERCURY	81 204.38 <b>Tl</b> THALLIUM	82 207.2 <b>Pb</b> LEAD	83 208.98 <b>Bi</b> BISMUTH	84 209 <b>Po</b> POLONIUM	85 210 <b>At</b> ASTATINE	86 222 <b>Rn</b> RADON																117 [...] <b>Uue</b> UNSEVEN	118 [...] <b>Uuo</b> EIGHT
87 [223] <b>Fr</b> FRANCIUM	88 [226] <b>Ra</b> RADIUM	89-103 Ac-Lr ACTINIDE	104 [261] <b>Rf</b> RUFORNIUM	105 [262] <b>Db</b> DUBNIUM	106 [271] <b>Sg</b> SEABORGIUM	107 [272] <b>Bh</b> BOHRVIUM	108 [273] <b>Hs</b> HASSIUM	109 [276] <b>Mt</b> MEITNERIUM	110 [281] <b>Ds</b> DARMSTADTIUM	111 [282] <b>Rg</b> ROENTGENIUM	112 [285] <b>Cn</b> COOPERSIUM	113 [...] <b>Uut</b> UNTRIUM	114 [287] <b>Fl</b> FLEROVIUM	115 [...] <b>Uup</b> UNPENTIUM	116 [291] <b>Lv</b> LIVERMORIUM	117 [...] <b>Uus</b> UNSEXTIUM	118 [...] <b>Uuo</b> UNOCTIUM																119 [...] <b>Uuh</b> UNUNNIUM	120 [...] <b>Uuq</b> UNBICHIUM

RELATIVE ATOMIC MASS (1)

GROUP 13

GROUP 14

ATOMIC NUMBER

SYMBOL

ELEMENT NAME

Metal

Semimetal

Nonmetal

Alkal metal

Alkaline earth metal

Transition metals

Lanthanide

Actinide

Chalcogens element

Halogens element

Noble gas

STANDARD STATE (25 °C, 101 kPa)

No - gas

Fe - solid

Hg - liquid

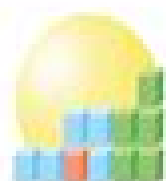
Ti - synthetic

## LANTHANIDE

57 138.91 <b>La</b> LANTHANUM	58 140.12 <b>Ce</b> CELIUM	59 140.91 <b>Pr</b> PRASEODYMIUM	60 144.24 <b>Nd</b> NEODYMIUM	61 [145] <b>Pm</b> PROMETHIUM	62 150.36 <b>Sm</b> SAMARIUM	63 151.96 <b>Eu</b> EUROPIUM	64 157.25 <b>Gd</b> GADOLINIUM	65 158.93 <b>Tb</b> TERBIUM	66 162.50 <b>Dy</b> DYSPROSIUM	67 164.93 <b>Ho</b> HOLMIUM	68 167.26 <b>Er</b> ERBIUM	69 168.93 <b>Tm</b> THULIUM	70 173.05 <b>Yb</b> YTERBIUM	71 174.97 <b>Lu</b> LUTETIUM
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## ACTINIDE

89 [227] <b>Ac</b> ACTINIUM	90 232.04 <b>Th</b> THORIUM	91 231.04 <b>Pa</b> PROTACTINIUM	92 238.03 <b>U</b> URANIUM	93 [237] <b>Np</b> NEPTUNIUM	94 [244] <b>Pu</b> PLUTONIUM	95 [247] <b>Am</b> AMERICIUM	96 [251] <b>Cm</b> CURIUM	97 [252] <b>Bk</b> BERKELEYIUM	98 [256] <b>Cf</b> CALIFORNIUM	99 [257] <b>Es</b> EINSTEINIUM	100 [261] <b>Fm</b> FERMIUM	101 [262] <b>Md</b> MEYERIUM	102 [265] <b>No</b> NORBOLIVIUM	103 [269] <b>Lr</b> LAWRENCIUM
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# Chapter 6 Classification Of Elements The Periodic Table

## Prentice Hall Exploring Physical Science

**D. Michael P. Mingos**



## **Chapter 6 Classification Of Elements The Periodic Table Prentice Hall Exploring Physical Science:**

**Prentice Hall Science Series, 1994** Prentice-Hall Staff,1993      **The Periodic Table** Eric R. Scerri,2006-10-12 The periodic table is one of the most potent icons in science It lies at the core of chemistry and embodies the most fundamental principles of the field The one definitive text on the development of the periodic table by van Spronsen 1969 has been out of print for a considerable time The present book provides a successor to van Spronsen but goes further in giving an evaluation of the extent to which modern physics has or has not explained the periodic system The book is written in a lively style to appeal to experts and interested lay persons alike The Periodic Table begins with an overview of the importance of the periodic table and of the elements and it examines the manner in which the term element has been interpreted by chemists and philosophers The book then turns to a systematic account of the early developments that led to the classification of the elements including the work of Lavoisier Boyle and Dalton and Cannizzaro The precursors to the periodic system like D bereiner and Gmelin are discussed In chapter 3 the discovery of the periodic system by six independent scientists is examined in detail Two chapters are devoted to the discoveries of Mendeleev the leading discoverer including his predictions of new elements and his accommodation of already existing elements Chapters 6 and 7 consider the impact of physics including the discoveries of radioactivity and isotopy and successive theories of the electron including Bohr s quantum theoretical approach Chapter 8 discusses the response to the new physical theories by chemists such as Lewis and Bury who were able to draw on detailed chemical knowledge to correct some of the early electronic configurations published by Bohr and others Chapter 9 provides a critical analysis of the extent to which modern quantum mechanics is or is not able to explain the periodic system from first principles Finally chapter 10 considers the way that the elements evolved following the Big Bang and in the interior of stars The book closes with an examination of further chemical aspects including lesser known trends within the periodic system such as the knight s move relationship and secondary periodicity as well at attempts to explain such trends      *Exploring the Elements* Isabel Thomas,2020 Science meets design in this comprehensive introduction to the chemical elements that make up our universe      **The Periodic Table** Sharon Katz Cooper,2007 An introduction to chemistry s periodic table of elements

*The Periodic Table* Sharon Katz Cooper,2007-07-31 Discusses Mendeleev s discovery known as the periodic table and shows how it has been instrumental to advancements in the field of chemistry

**The Periodic Table** Eric R. Scerri,2019 Eric R Scerri presents a modern and fresh exploration of this fundamental topic in the physical sciences considering the deeper implications of the arrangements of the table to atomic physics and quantum mechanics This new edition celebrates the completion of the 7th period of the table with the naming of elements 113 115 117 and 118      *The World of Elements and Their Properties* Lynn Van Gorp,2007-08-03 From compounds to chemical reactions readers will learn all about elements their properties and how they react with other elements in this stunning book that features colorful images and intriguing facts Ionic bonds chemical bonds the Periodic Table of Elements mixtures and

solutions are some of the topics that are discussed The accessible glossary and index gives readers the tools they need to better understand the content while a fascinating hands on lab activity will leave readers engaged and excited to learn more

Elements and the Periodic Table Suzanne Slade,2007-08-01 Examines elements that people have studied throughout history including the periodic table      **Elements and the Periodic Table, Grades 5 - 12** Theodore A. Abbgy,2001-01-01 This informative classroom supplement is a great introduction to the periodic table explored in sequential form It includes activities transparency masters a teacher s guide an element game quizzes tests rubrics and answer keys Unit topics include discovering what elements are the uses of the elements element symbols periodic table organization and more Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources      The Periodic Table Adrian Dingle,Simon Basher,2007 A mixture of science and art creates a periodic table never seen before      A Guide to the Elements Albert Stwertka,2012-02-01 Newly updated throughout and now covering 118 elements this crystal clear guide to the periodic table illuminates the basic concepts of chemistry as it traces the history and development of our knowledge of the material world In this fascinating volume Albert Stwertka makes complex ideas and terms easily understandable drawing upon engaging historical anecdotes and everyday examples to clarify the text which is complemented by numerous illustrations many in full color Since the second edition many new elements have been named and discovered including Darmstadtium Roentgenium and Copernicium and the elements currently called Ununtrium Ununpentium Ununhexium Ununseptium and Ununoctium The third edition provides thorough coverage of all these new discoveries In addition to the new elements Stwertka has brought the information about the elements in the second edition up to date based on the latest research He discusses a cylindrical molecule of carbon known as a nanotube which has become a do all wonder substance touted for use in everything from X ray machines to paint A new form of the element boron has been found that is nearly as hard as diamond Its superior heat resistance could make it attractive for certain industrial uses And a new particle detector using ultra pure liquid xenon has been constructed beneath 5 000 feet of rock in Italy to detect dark matter Stwertka also covers the 2010 Nobel winning work on graphene an ultrathin form of carbon that is vital for future generations of computers and touch screens the discovery of new superconductors and the development of new uses for the rare earth elements Bringing the periodic table into the 21st century this engrossing guide to the elements will fascinate everyone curious about the basic building blocks of the material world      **The Periodic Table I** D. Michael P. Mingos,2020-02-05 As 2019 has been declared the International Year of the Periodic Table it is appropriate that Structure and Bonding marks this anniversary with two

special volumes In 1869 Dmitri Ivanovitch Mendeleev first proposed his periodic table of the elements He is given the major credit for proposing the conceptual framework used by chemists to systematically inter relate the chemical properties of the elements However the concept of periodicity evolved in distinct stages and was the culmination of work by other chemists over several decades For example Newland s Law of Octaves marked an important step in the evolution of the periodic system since it represented the first clear statement that the properties of the elements repeated after intervals of 8 Mendeleev s predictions demonstrated in an impressive manner how the periodic table could be used to predict the occurrence and properties of new elements Not all of his many predictions proved to be valid but the discovery of scandium gallium and germanium represented sufficient vindication of its utility and they cemented its enduring influence Mendeleev s periodic table was based on the atomic weights of the elements and it was another 50 years before Moseley established that it was the atomic number of the elements that was the fundamental parameter and this led to the prediction of further elements Some have suggested that the periodic table is one of the most fruitful ideas in modern science and that it is comparable to Darwin s theory of evolution by natural selection proposed at approximately the same time There is no doubt that the periodic table occupies a central position in chemistry In its modern form it is reproduced in most undergraduate inorganic textbooks and is present in almost every chemistry lecture room and classroom This first volume provides chemists with an account of the historical development of the Periodic Table and an overview of how the Periodic Table has evolved over the last 150 years It also illustrates how it has guided the research programmes of some distinguished chemists

*Basher Science: The Complete Periodic Table* Adrian Dingle, Simon Basher, Dan Green, 2015-01-06 Combines science and art to create a simple way for students to learn the periodic table

**Dial My Number! What is an Element's Atomic Number, Symbol and Mass Number | Periodic Table | Grade 6-8 Physical Science** Dot EDU, 2024-04-15 Students will learn about elements with a crucial guide for grades 6 8 emphasizing the STEM curriculum s core Dial My Number explores atomic numbers symbols and mass numbers providing a fundamental understanding of the Periodic Table From the structure of atoms to the intriguing process of isotopes and the significance of element symbols this book is an essential resource for teachers homeschooling parents and librarians Ensure your students or children grasp these foundational concepts in Physical Science by adding this insightful book to your educational tools

**The Elements: A Very Short Introduction** Philip Ball, 2004-04-08 This Very Short Introduction is an exciting and non traditional approach to understanding the terminology properties and classification of chemical elements It traces the history and cultural impact of the elements on humankind from ancient times through today Packed with anecdotes The Elements is a highly engaging and entertaining exploration of the fundamental question what is the world made from

**How to Read the Periodic Table of Elements | Chemistry for Beginners Grade 5 | Children's Science & Nature Books** Baby Professor, 2021-11-01 At the end of this book your child should be able to read and understand some of the periodic table He she should gain the basic understanding

of how elements can be identified in symbols and that the table can be read from left to right Moreover your child will also learn what atomic numbers mean and how the table is divided between elements Start reading today

**A Tale of Seven Elements** Eric Scerri, 2013-07-18 In A Tale of Seven Elements Eric Scerri presents the fascinating history of those seven elements discovered to be mysteriously missing from the periodic table in 1913

[The Mathematics of the Periodic Table](#) D. H. Rouvray, R. Bruce King, 2006 The Periodic Table effectively embraces the whole realm of chemistry within the confines of one comparatively simple and easily understood chart of the chemical elements Over many years the Periodic Table has proven to be indispensable not only to chemists of all kinds but also to a host of other scientists including biologists geologists and physicists It is thus hardly surprising that the Periodic Table has become one of our most celebrated contemporary scientific icons In the present work various aspects of the Periodic Table that are seldom if ever featured elsewhere are given prominence The twelve presentations contained herein all have a mathematical flavour because it is the intention to highlight the often neglected mathematical features of the Periodic Table and several closely related topics The book starts out by considering predictions of what the ultimate size of the Periodic Table will be when all of the possible artificial chemical elements have been synthesised It then moves on to an examination of the nature of the periodicity extant in the Periodic Table and some methods for the prediction of the properties of the super heavy elements The Periodic Table is next explored in various dimensions other than two The natural clustering of the elements into groups is studied by three different but complementary routes namely via the topological structures of the groups the self association of the elements as evidenced by neural network studies and information theoretical analysis of the behaviour of atoms Following a detailed investigation of the mathematical basis for the periodicity seen in atomic and molecular spectroscopy three separate presentations delve into many different aspects of the group theoretical structure of the Periodic Table The unusual combination of themes offered here will appeal to all who seek a more detailed and intimate knowledge of the Periodic Table than that available in standard texts on the subject

**The Lost Elements** Marco Fontani, Mariagrazia Costa, Mary Virginia Orna, 2014-10-01 The Periodic Table of Elements hasn't always looked like it does now a well organized chart arranged by atomic number In the mid nineteenth century chemists were of the belief that the elements should be sorted by atomic weight However the weights of many elements were calculated incorrectly and over time it became clear that not only did the elements need rearranging but that the periodic table contained many gaps and omissions there were elements yet to be discovered and the allure of finding one had scientists rushing to fill in the blanks Supposed discoveries flooded laboratories and the debate over what did and did not belong on the periodic table reached a fever pitch With the discovery of radioactivity the discourse only intensified Throughout its formation the Periodic Table of Elements has seen false entries good faith errors retractions and dead ends In fact there have been more falsely proclaimed elemental discoveries throughout history than there are elements on the table as we know it today

The Lost Elements The Periodic Table's Shadow

Side collects the most notable of these instances stretching from the nineteenth century to the present. The book tells the story of how scientists have come to understand elements by discussing the failed theories and false discoveries that shaped the path of scientific progress. We learn of early chemists' stubborn refusal to disregard alchemy as a legitimate practice and of one German's supposed discovery of an elemental metal that breathed. As elements began to be created artificially in the twentieth century, we watch the discovery climate shift to favor the physicists rather than the chemists. Along the way, Fontana, Costa, and Orna introduce us to the key figures in the development of today's periodic table, including Lavoisier and Mendeleev. Featuring a preface from Nobel Laureate Roald Hoffmann, *The Lost Elements* is an expansive history of the wrong side of chemical discovery and reveals how these errors and gaffes have helped shape the table as much as any other form of scientific progress.

The Periodic Table Becky Ham, 2009. A short guide to one of the most celebrated diagrams in the history of science, a tool for understanding the basic building blocks of the universe.

## **Chapter 6 Classification Of Elements The Periodic Table Prentice Hall Exploring Physical Science** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Chapter 6 Classification Of Elements The Periodic Table Prentice Hall Exploring Physical Science**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

<https://www.portal.goodeyes.com/data/scholarship/Documents/Eine%20Frau%20Reist%20Durch%20Welt%20Ebook.pdf>

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### **Chapter 6 Classification Of Elements The Periodic Table Prentice Hall Exploring Physical Science Introduction**

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