

**CHAPTER 19 STUDY GUIDE FOR CONTENT MASTERY**

# Acids and Bases

## Section 19.1 Acids and Bases: An Introduction

In your notebook, read about the properties of acids and bases.

For each description below, write *acid* if it tells about a property of an acid or *base* if it tells about a property of a base. If the property does not apply to either an acid or a base, write *neither*. If it applies to both an acid and a base, write *both*.

1. Can turn litmus paper a different color both
2. Reacts with certain metals acid
3. Contains more hydrogen ions than hydroxide ions acid
4. Feels slippery base
5. Reacts with carbonates acid
6. Feels rough neither
7. Contains equal numbers of hydrogen and hydroxide ions neither
8. Tastes bitter base
9. Tastes sour acid

In your notebook, read about the different models of acids and bases.

Use the terms below to complete the passages. You may use each term more than once.

Arrhenius conjugate base	Brønsted-Lowry hydrogen	conjugate acid hydroxide
-----------------------------	----------------------------	-----------------------------

The (10) Arrhenius model of acids and bases states that an acid contains the element (11) hydrogen and forms ions of this element when it is dissolved in water. A base contains the (12) hydroxide group and dissociates to produce (13) hydroxide ions in aqueous solution.

According to the (14) Brønsted-Lowry model, an acid donates

(15) hydrogen ions, and a base accepts (16) hydrogen ions.

According to this model, in an acid-base reaction, each acid has a

(17) conjugate base and each base has a (18) conjugate acid.

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## Section 19.2 Strengths of Acids and Bases

In your notebook, read about strengths of acids.

Circle the letter of the choice that best completes the statement or answers the question.

1. Acid A and acid B are of equal concentration and are tested with a conductivity apparatus. When the electrodes are placed in acid A, the bulb glows dimly. When they are placed in acid B, the bulb glows more brightly. Which of the following is true?
  - a. Acid A is stronger than acid B.
  - b. Acid B is stronger than acid A.
  - c. Acid A and acid B are of equal strength.
  - d. No comparison of strength can be made from the results.
2. A chemical equation for the ionization of an acid runs a single arrow to the right ( $\rightarrow$ ) to separate the reactant and product sides of the equation. Which of the following is true?
  - a. The arrow does not indicate relative strength.
  - b. The ionizing acid is strong.
  - c. The ionizing acid is half ionized.
  - d. The ionizing acid is weak.
3. Sulfuric acid is a strong acid. What is true about its conjugate base?
  - a. Its conjugate base is amphiprotic.
  - b. Its conjugate base is strong.
  - c. Its conjugate base is weak.
  - d. No conclusion can be made regarding the strength of the conjugate base.
4. In solution, a weak acid produces
  - a. a mixture of molecules and ions.
  - b. all ions.
  - c. all molecules.
  - d. anions, but no hydroxide ions.
5. Why are  $K_a$  values of weak numbers?
  - a. The concentration of water does not affect the ionization.
  - b. The equilibrium is not stable.
  - c. The solution contains a high concentration of ions.
  - d. The solution contains a high concentration of un-ionized acid molecules.
6. Which of the following dissociates nearly one hundred percent in hydroxide ions in solution?
  - a. a strong acid
  - b. a strong base
  - c. a weak acid
  - d. a weak base
7. In general, compounds formed from active metals and hydroxide ions are
  - a. strong acids
  - b. strong bases
  - c. weak acids
  - d. weak bases

# Chapter 19 Study Guide For Content Mastery Answers

## Acids And Bases

**Sir James Augustus Henry Murray**



## Chapter 19 Study Guide For Content Mastery Answers Acids And Bases:

**Fundamentals of Biochemistry** Donald Voet, Judith G. Voet, Charlotte W. Pratt, 2016-02-29 Voet Voet and Pratt's Fundamentals of Biochemistry 5th Edition addresses the enormous advances in biochemistry particularly in the areas of structural biology and Bioinformatics by providing a solid biochemical foundation that is rooted in chemistry to prepare students for the scientific challenges of the future While continuing in its tradition of presenting complete and balanced coverage that is clearly written and relevant to human health and disease Fundamentals of Biochemistry 5e includes new pedagogy and enhanced visuals that provide a pathway for student learning Resources in education, 1988 *Biology* Sylvia S. Mader, 2000-07 *Resources in Education*, 1984 **Basic Nursing** Patricia Ann Potter, Anne Griffin Perry, 1991 *Indexes* United States. Environmental Protection Agency, 1983 Cumulated Index Medicus, 1968 Approach to Chemistry University of New South Wales. Summer School for Chemistry Teachers, 1962 *Approach to Chemistry*, 1962 Bulletin of the Atomic Scientists, 1972-10 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world *Popular Mechanics*, 2000-01 Popular Mechanics inspires instructs and influences readers to help them master the modern world Whether it's practical DIY home improvement tips gadgets and digital technology information on the newest cars or the latest breakthroughs in science PM is the ultimate guide to our high tech lifestyle **Los Angeles Magazine**, 2003-11 Los Angeles magazine is a regional magazine of national stature Our combination of award winning feature writing investigative reporting service journalism and design covers the people lifestyle culture entertainment fashion art and architecture and news that define Southern California Started in the spring of 1961 Los Angeles magazine has been addressing the needs and interests of our region for 48 years The magazine continues to be the definitive resource for an affluent population that is intensely interested in a lifestyle that is uniquely Southern Californian Bulletin of the Atomic Scientists, 1967-03 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world **The Continuing Study of Newspaper Reading** Advertising Research Foundation, 1942 *EPA Publications Bibliography*, 1977 Government Reports Announcements & Index, 1980-11 Government Reports Annual Index, 1983 **The Compact Edition of the Oxford English Dictionary** Sir James Augustus Henry Murray, 1971 Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933

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