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# 17

## THERMOCHEMISTRY

### SECTION 17.1 THE FLOW OF ENERGY—HEAT AND WORK (pages 505–510)

*This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat.*

#### ► Energy Transformations (page 505)

1. What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions? thermochemistry
2. Where the use of energy is concerned (in a scientific sense), when is work done?  
Work is done when a force is used to move an object.
3. Circle the letter next to each sentence that is true about energy:  
☒ a. Energy is the capacity for doing work or supplying heat.  
☒ b. Energy is detected only because of its effects.  
☐ c. Heat is energy that transfers from one object to another because they are at the same temperature.  
☒ d. Gasoline contains a significant amount of chemical potential energy.
4. Circle the letter next to each sentence that is true about heat.  
☒ a. One effect of adding heat to a substance is an increase in the temperature of that substance.  
☐ b. Heat always flows from a cooler object to a warmer object.  
☒ c. If two objects remain in contact, heat will flow from the warmer object to the cooler object until the temperature of both objects is the same.

#### ► Exothermic and Endothermic Processes (pages 506–507)

5. What can be considered the “system” and what are the “surroundings” when studying a mixture of chemicals undergoing a reaction? Write your answers where indicated below.

System: The mixture of chemicals itself is considered the system.

Surroundings: Everything but the mixture of chemicals is the surroundings, but practically speaking, the immediate vicinity of the system.

# Chapter 171 The Flow Of Energy Heat And Work Answers

**Wolfgang Guggemos**



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## **Table of Contents Chapter 171 The Flow Of Energy Heat And Work Answers**

1. Understanding the eBook Chapter 171 The Flow Of Energy Heat And Work Answers
  - The Rise of Digital Reading Chapter 171 The Flow Of Energy Heat And Work Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 171 The Flow Of Energy Heat And Work Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 171 The Flow Of Energy Heat And Work Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 171 The Flow Of Energy Heat And Work Answers
  - Personalized Recommendations
  - Chapter 171 The Flow Of Energy Heat And Work Answers User Reviews and Ratings
  - Chapter 171 The Flow Of Energy Heat And Work Answers and Bestseller Lists
5. Accessing Chapter 171 The Flow Of Energy Heat And Work Answers Free and Paid eBooks
  - Chapter 171 The Flow Of Energy Heat And Work Answers Public Domain eBooks
  - Chapter 171 The Flow Of Energy Heat And Work Answers eBook Subscription Services
  - Chapter 171 The Flow Of Energy Heat And Work Answers Budget-Friendly Options
6. Navigating Chapter 171 The Flow Of Energy Heat And Work Answers eBook Formats

- ePub, PDF, MOBI, and More
  - Chapter 171 The Flow Of Energy Heat And Work Answers Compatibility with Devices
  - Chapter 171 The Flow Of Energy Heat And Work Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chapter 171 The Flow Of Energy Heat And Work Answers
    - Highlighting and Note-Taking Chapter 171 The Flow Of Energy Heat And Work Answers
    - Interactive Elements Chapter 171 The Flow Of Energy Heat And Work Answers
  8. Staying Engaged with Chapter 171 The Flow Of Energy Heat And Work Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chapter 171 The Flow Of Energy Heat And Work Answers
  9. Balancing eBooks and Physical Books Chapter 171 The Flow Of Energy Heat And Work Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chapter 171 The Flow Of Energy Heat And Work Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chapter 171 The Flow Of Energy Heat And Work Answers
    - Setting Reading Goals Chapter 171 The Flow Of Energy Heat And Work Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chapter 171 The Flow Of Energy Heat And Work Answers
    - Fact-Checking eBook Content of Chapter 171 The Flow Of Energy Heat And Work Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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