

Solutions

Reviewing Vocabulary

Match the definition in Column A with the terms in Column B.

Column A

- _____ 1. The diffusion of solvent particles across a semipermeable membrane from areas of lower solute concentration to areas of higher solute concentration
- _____ 2. A mixture with particles that settle out if undisturbed
- _____ 3. The erratic movement of colloid particles
- _____ 4. The amount of additional pressure caused by water molecules moving into a solution
- _____ 5. A measure of how much solute is dissolved in a specific amount of solvent or solution
- _____ 6. The overall energy change that occurs when a solution forms
- _____ 7. A heterogeneous mixture of intermediate size particles
- _____ 8. The process of surrounding solute particles with solvent particles to form a solution
- _____ 9. The ratio of the number of moles of solute in solution to the total number of moles of solute and solvent
- _____ 10. The scattering of light by dispersed colloid particles
- _____ 11. The statement that the solubility of a gas in a liquid is directly proportional to the pressure of the gas above the liquid

Column B

- a. Brownian motion
- b. colloid
- c. concentration
- d. heat of solution
- e. Henry's law
- f. mole fraction
- g. osmosis
- h. osmotic pressure
- i. solvation
- j. Tyndall effect
- k. suspension

Describe each pair of related terms.

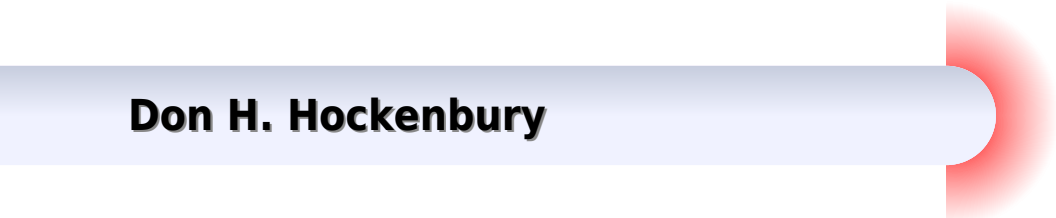
12. soluble, insoluble

13. miscible, immiscible

14. molarity, molality

Chapter 15 Study Guide

Don H. Hockenbury



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