

States of Matter

Multiple-Choice Questions

Answer questions 1-10 about the topics mentioned below.

Complete each statement.

1. The kinetic molecular theory describes the behavior of particles when in gaseous state.
2. The kinetic molecular theory states the following assumptions:
 - a. It is assumed gas particles have no attractive forces between any _____ (except for the volume of the particle).
 - b. Because gas particles are in constant random significant motion, _____, therefore gas particles _____.
 - c. Gas particles in contact with _____ exert _____ of the collisions between gas particles _____ that have _____ change in _____.
3. The kinetic theory of gases is _____ the liquid.
4. _____ is a common solid property where the particles are packed at least.

Answer questions 11-15 about properties of different states of matter.

Fill in the answers below, whether or not.

11. _____ are the three states of matter that are in a solid state, liquid state, and gas.
12. The _____ and particles move a gas is _____ in the _____.
13. The _____ of a gas is _____.
14. _____ are _____ compared to _____.
15. The _____ is _____ by the _____ of the particles of the gas.
16. _____ gas particles _____ than in _____ gas particles.
17. _____ is _____ through _____, _____.
18. _____ is _____ that _____ of _____ but a gas is _____, _____ of the _____ of its particles.

Answer questions 16-20 about properties.

Complete the table that shows the three properties that are common to all states of matter.

16. _____ the gas.

(a) _____

(b) _____

(c) _____

(d) _____

17. _____ the gas.

(a) _____

(b) _____

(c) _____

(d) _____

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SJ Ball



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