

## Practice

- 1)  $\text{TiO}_2 + \text{C} + 2\text{Cl}_2 \rightarrow \text{TiCl}_4 + \text{CO}_2$ 
  - If you begin with 1.25 mol  $\text{TiO}_2$ , what mass of  $\text{Cl}_2$  gas is needed?



# Chapter 12 Study Guide Chemistry Stoichiometry

## Answer Key

**D Kirk**



## **Chapter 12 Study Guide Chemistry Stoichiometry Answer Key:**

## Whispering the Strategies of Language: An Emotional Journey through **Chapter 12 Study Guide Chemistry Stoichiometry Answer Key**

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