

Lab Data

Freezing-point depression (°C)

Select

Freezing-point depression constant (°C/ <i>m</i> )		1.86	
Freezing-point of DI water (°C)		0.10	
	Solution 1	Solution 2	Solution 3
Magnesium chloride	MgCl <sub>2</sub>	MgCl <sub>2</sub>	MgCl <sub>2</sub>
Molar mass (g/mol)	95.21	95.21	95.21
Mass of calorimeter (g)	17.22	17.45	17.33
Volume of DI water (mL)	47.3	48.3	49.0
Mass of magnesium chloride (g)	1.58	2.15	3.09
Moles of magnesium chloride (mol)	0.0166	0.0226	0.0325
Mass of calorimeter + solution (g)	66.12	67.78	69.53
Mass of calorimeter + solution + ice (g)	110.45	113.96	114.54
Freezing-point of solution (°C)	-1.12	-1.30	-1.96
Mass of calorimeter + ice after decanting (g)	39.88	30.37	31.62
Mass of ice added (g)	44.33	46.18	45.01
Mass of ice remaining after decanting (g)	22.66	12.92	14.29
Mass of melted ice (g)	21.67	33.26	30.72
Mass of water in final solution (g)	68.99	81.44	79.83
Mass of water in final solution (kg)	0.06899	0.08144	0.07983
Molality of solution ( <i>m</i> )	0.241	0.278	0.407
Freezing-point depression (°C)	1.22	1.40	2.06
van't Hoff factor			



[How to calculate van't Hoff factor](#)

Lines:

RESE

# Colligative Properties Freezing Point Depression Lab Answers

**LL Leslie**

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the LL Leslie text.

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