

Sample Placement Exam for Chemistry 200

The Placement Exam contains 44 multiple-choice questions. You will be allowed 45 minutes to do the exam. A good workbook for reviewing for the test is "Schaum's Easy Outlines, Beginning Chemistry". Silent, non-printing, non-programmable calculators will be permitted; Personal Digital Assistants will not be permitted. Your score will be the total of correct answers. A periodic table and other useful information will be provided on the back page. Scratch paper will also be provided.

The exam covers the following topics in chemistry and mathematics at the high school level.
Elements and Atoms: basic properties, atomic number, atomic mass, isotopes, Lewis symbols, valence electrons, electronic configurations, use of the periodic table

Compounds and Molecules: physical properties, molecular mass, naming inorganic compounds, formula calculations, simple Lewis dot structures

Reactions: Balancing equations, simple stoichiometry, equilibria, enthalpy, entropy, simple kinetics

Bonding: Ionic, covalent, metallic

Gases: Use of the ideal gas law

Solutions: Molarity, dilution, basic ideas of acids and bases, properties of electrolytes

Mathematics: significant figures, simple algebra

The following twelve questions are examples of the types of questions that may be asked, but note that they do not cover all of the material that may be on the exam.

Circle the one best answer for each of the following questions.

1. What is the molar mass (molecular weight) of the compound Al_2O_3 ?
A. 21.00
B. 42.98
C. 50.00
D. 101.96
2. The density of ethanol is 0.789 g/cm^3 . How many grams are there in 205 milliliters of ethanol?
A. 162
B. 205
C. 260
D. 2.60×10^3
3. Which chemical equation below is *balanced*?
A. $2 \text{N}_2(\text{g}) + 3 \text{H}_2(\text{g}) \rightarrow 2 \text{NH}_3(\text{g})$
B. $2 \text{Mg}(\text{s}) + \text{O}_2(\text{g}) \rightarrow 2 \text{MgO}(\text{s})$
C. $2 \text{Al}(\text{s}) + 3 \text{O}_2(\text{g}) \rightarrow \text{Al}_2\text{O}_3(\text{s})$
D. $\text{Ca}[\text{OH}]_2(\text{s}) + \text{CO}_2(\text{g}) \rightarrow \text{Ca}[\text{HCO}_3]_2(\text{s})$
4. How many grams of sugar are in six dozen doughnuts if each doughnut contains 75 g of sugar?
A. 0.96
B. 450
C. 900
D. 5400

College Chemistry Placement Test Study Guide

Richard O Peterson



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AP Chemistry, All Access Kevin Reel, Derrick C. Wood, Scott A. Best, 2014 All Access for the AP R Chemistry Exam Completely Revised for the 2014 Exam Book Web Mobile Everything you need to prepare for the Advanced Placement R exam in a study system built around you There are many different ways to prepare for an Advanced Placement R exam What s best for you depends on how much time you have to study and how comfortable you are with the subject matter To score your highest you need a system that can be customized to fit you your schedule your learning style and your current level of knowledge This book and the online tools that come with it will help you personalize your AP R Chemistry prep by testing your understanding pinpointing your weaknesses and delivering flashcard study materials unique to you The REA AP R All Access system allows you to create a personalized study plan through three simple steps targeted review of exam content assessment of your knowledge and focused study in the topics where you need the most help Here s how it works Review the Book Study the Big Ideas tested on the AP R Chemistry exam and learn proven AP R strategies that will help you tackle any question you may see on test day Test Yourself and Get Feedback As you review the book test yourself with 7 end of chapter quizzes plus two mini tests Score reports from your free online tests and quizzes give you a fast way to pinpoint what you really know and what you should spend more time studying Improve Your Score Armed with your score reports you can personalize your study plan Review the parts of the book where you are weakest and use the REA Study Center to create your own unique AP R Chemistry e flashcards adding to the 100 free cards included with this book Visit The REA Study Center for a suite of online tools The best way to personalize your study plan is to get frequent feedback on what you know and what you don t At the online REA Study Center you can access three types of assessment topic level quizzes mini tests and a full length practice test Each of these tools provides true to format questions and delivers a detailed score report that follows the topics set by the College Board R Topic Level Quizzes Short 15 minute online quizzes are available throughout the review and are designed to test your immediate grasp of the topics just covered Mini Tests Two online mini tests cover what you ve studied in each half of the book These tests are like the actual AP R Chemistry exam only shorter and will help you evaluate your overall understanding of the subject Full Length Practice Test After you ve finished reviewing the book take our full length AP R Chemistry exam to practice under test day conditions Available both in the book and online this test gives you the most complete picture of your strengths and weaknesses We strongly recommend that you take the online version of the exam for the added benefits of timed testing automatic scoring and a detailed score report Improving Your Score e Flashcards With your score reports from the quizzes and tests you ll be able to see exactly which AP R Chemistry topics you need to review Use this information to create your own flashcards for the areas where you are weak And because you will create these flashcards through the REA Study Center you ll be able to access them from any computer or smartphone Not quite sure what to put on your flashcards Start with the 100 free cards included when you buy this book This

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