

Student Name: \_\_\_\_\_

Student ID: \_\_\_\_\_

# **CHEM 2218 Inorganic Chemistry I (Final Exam sample paper)**

## **Part I: Multiple Choices** (2 points each, 50 points in total)

- For a cube, the number of  $C_6$ ,  $C_4$ , and  $C_2$  rotational axes is  
(a) 1, 3, and 4, respectively (b) 0, 2, and 4, respectively (c) 0, 3, and 4, respectively
- The CO diatomic molecule contains infinite number of  
(a)  $\sigma_v$  (b)  $i$  (c)  $\sigma_h$  (d)  $C_2$
- Identify the wrong label(s) in the following list used for representations of point groups.  
 $C_{1000}$ ,  $C_{1000}$ ,  $D_{1000}$ ,  $D_{2v}$ ,  $C_4$ ,  $I$ ,  $O_h$ ,  $T_d$
- The  $Fe_2(CO)_9$  molecule shown does not possess what element of symmetry?  
(a)  $i$  (b)  $\sigma$  (c)  $S_6$  (d)  $C_2$
- A molecule having a point group of  $D_{6h}$  contains all the symmetry elements of a molecule having a point group of  
(a)  $D_{4d}$  (b)  $D_{2d}$  (c)  $D_{3h}$  (d) none of the above
- How many mirror planes of symmetry are there in an icosahedron?  
(a) 6 (b) 9 (c) 12 (d) 15 (e) 18
- Give the symmetry label for the molecular orbital associated with a trigonal planar structure (*you are given a character table for  $D_{3h}$* ).  
(a)  $A_1'$  (b)  $A_1''$   
(c)  $A_2'$  (d)  $A_2''$
- Identify the stronger acid?  
 $H_2Se$  vs.  $H_2S$        $[Al(H_2O)_6]^{3+}$  vs.  $[Fe(H_2O)_6]^{3+}$
- Which one in the following is the least stable?  
(a)  $[Al(H_2O)_6]^{3+}$  (b)  $[AlF_6]^{3-}$  (c)  $Mo(H_2O)_6$  (d)  $Mo(CO)_6$
- Predict which way the following reactions will go (left or right)  
(a)  $[AlF_6]^{3-} + Ga(CN)_6^{3-} \rightarrow [GaF_6]^{3-} + Al(CN)_6^{3-}$   
(a)  $W(CO)_5(PMe_3) + NMe_3 \rightarrow W(CO)_5(NMe_3) + PMe_3$
- Select the strongest base toward proton.  
(a)  $NH_3$  (b)  $NHMe_2$  (c)  $PH_3$  (d)  $PMe_3$
- Which of the following reactions belong to the acid-base reactions?  
(a)  $H + OH \rightarrow H_2O$ .  
(b)  $PF_3 + F_2 \rightarrow PF_5$ .  
(c)  $SiF_4 + 2NaF \rightarrow Na_2[SiF_6]$ .  
(d)  $Mo + 6CO \rightarrow Mo(CO)_6$ .



# College Chemistry 1 Final Exam Practice Questions

**JL Elias**



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caused by being anxious Work on the easier questions first If you find yourself working too long on one question make a mark next to it in your test booklet and continue After you have answered all of the questions that you can go back to the ones you have skipped Use the process of elimination when you are unsure of an answer If you can eliminate three of the answer choices you have given yourself a fifty fifty chance of getting the item correct since there will only be two choices left from which to make a guess If you cannot eliminate at least three of the answer choices you may choose not to guess as you will be penalized one quarter of a point for every incorrect answer Questions not answered will not be counted Be sure that you are marking your answer in the oval that corresponds with the correct item in the test booklet Since the multiple choice section is graded by machine marking the wrong answer will throw off your score      *Assessing and Improving Your Teaching* Phyllis Blumberg,2013-09-06 In order to make appropriate changes to improve your teaching and your students learning first you need to know how you re teaching now Figure it out for yourself and invigorate your teaching on your own terms This practical evidence based guide promotes excellence in teaching and improved student learning through self reflection and self assessment of one s teaching Phyllis Blumberg starts by reviewing the current approaches to instructor evaluation and describes their inadequacies She then presents a new model of assessing teaching that builds upon a broader base of evidence and sources of support This new model leads to self assessment rubrics which are available for download and the book will guide you in how to use them The book includes case studies of completed critical reflection rubrics from a variety of disciplines including the performing and visual arts and the hard sciences to show how they can be used in different ways and how to explore the richness of the data you ll uncover      *The College Professor as the Student Sees Him* Hermann Henry Remmers,1929      **Resources in Education** ,1996-05      *Fuel* Joe White,2012-07-13 Finding ways to connect on a spiritual level with teens can be difficult With these simple 10 minute devotionals parents can maximize their devotional time with their teens and prepare and equip them with the strong spiritual foundation they need Written by youth expert Joe White Fuel makes it easy and practical for parents to connect spiritually with their teens in just minutes a day      **Graduate Programs in the Biological/Biomed Sciences & Health-Related/Med Prof 2015 (Grad 3)** Peterson's,2014-12-16 Peterson s Graduate Programs in the Biological Biomedical Sciences Health Related Medical Professions 2015 contains profiles of 6 750 graduate programs at over 1 200 institutions in the biological biomedical sciences and health related medical professions Informative data profiles are included for 6 750 graduate programs in every available discipline in the biological and biomedical sciences and health related medical professions including facts and figures on accreditation degree requirements application deadlines and contact information financial support faculty and student body profiles Two page in depth descriptions written by featured institutions offer complete details on specific graduate program school or department as well as information on faculty research and the college or university Comprehensive directories list programs in this volume as well as others in the graduate series      *Competition Science Vision* ,2001-08 Competition Science Vision monthly

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