

CHAPTER 15 STUDY GUIDE

Energy and Chemical Change

Section 15.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write *true* if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- | | |
|-------|---|
| _____ | 1. <i>Energy</i> is the ability to do work or produce heat. |
| _____ | 2. The law of conservation of energy states that energy <i>can be</i> created and destroyed. |
| _____ | 3. Chemical potential energy is energy stored in a substance because of its <i>composition</i> . |
| _____ | 4. <i>Heat</i> is a form of energy that flows from a warmer object to a cooler object. |
| _____ | 5. A calorie is the amount of energy required to raise the temperature of <i>one gram</i> of pure water by one degree Celsius. |
| _____ | 6. A <i>calorie</i> is the SI unit of heat and energy. |
| _____ | 7. The <i>specific heat</i> of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius. |
| _____ | 8. <i>Kinetic energy</i> is energy of motion. |
| _____ | 9. Chemicals participating in a chemical reaction contain only <i>potential energy</i> . |
| _____ | 10. One nutritional Calorie is equal to <i>100 calories</i> . |
| _____ | 11. One calorie equals <i>4.184 joules</i> . |
| _____ | 12. When a fuel is burned, some of its <i>chemical potential energy</i> is lost as heat. |
| _____ | 13. To convert kilojoules to joules, <i>divide</i> the number of kilojoules by 1000 joules/1 kilojoule. |

Answer the following question. Show all your work.

14. If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g·°C).

Energy And Chemical Study Guide Key

Xiaolong Qi



Energy And Chemical Study Guide Key:

The book delves into Energy And Chemical Study Guide Key. Energy And Chemical Study Guide Key is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Energy And Chemical Study Guide Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Energy And Chemical Study Guide Key
 - Chapter 2: Essential Elements of Energy And Chemical Study Guide Key
 - Chapter 3: Energy And Chemical Study Guide Key in Everyday Life
 - Chapter 4: Energy And Chemical Study Guide Key in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Energy And Chemical Study Guide Key. The first chapter will explore what Energy And Chemical Study Guide Key is, why Energy And Chemical Study Guide Key is vital, and how to effectively learn about Energy And Chemical Study Guide Key.
 3. In chapter 2, this book will delve into the foundational concepts of Energy And Chemical Study Guide Key. The second chapter will elucidate the essential principles that must be understood to grasp Energy And Chemical Study Guide Key in its entirety.
 4. In chapter 3, the author will examine the practical applications of Energy And Chemical Study Guide Key in daily life. The third chapter will showcase real-world examples of how Energy And Chemical Study Guide Key can be effectively utilized in everyday scenarios.
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 6. In chapter 5, this book will draw a conclusion about Energy And Chemical Study Guide Key. The final chapter will summarize the key points that have been discussed throughout the book.
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