

# History Chemical Analysis

## 12

### Question 10

What is the correct chemical formula for the compound formed by the reaction of iron(II) and sulfur?

FeS  
Fe<sub>2</sub>S<sub>3</sub>  
Fe<sub>3</sub>S<sub>2</sub>  
FeS<sub>2</sub>

### Answer

The correct answer is FeS. Iron(II) has a charge of +2 and sulfur has a charge of -2. The charges cancel out, resulting in the formula FeS.

What is the correct chemical formula for the compound formed by the reaction of iron(III) and sulfur?

- FeS
- Fe<sub>2</sub>S<sub>3</sub>
- FeS<sub>2</sub>
- Fe<sub>3</sub>S<sub>2</sub>

The correct answer is Fe<sub>2</sub>S<sub>3</sub>. Iron(III) has a charge of +3 and sulfur has a charge of -2. The charges cancel out, resulting in the formula Fe<sub>2</sub>S<sub>3</sub>.

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- FeS
- Fe<sub>2</sub>S<sub>3</sub>
- FeS<sub>2</sub>
- Fe<sub>3</sub>S<sub>2</sub>

What is the correct chemical formula for the compound formed by the reaction of iron(II) and sulfur?

- FeS
- Fe<sub>2</sub>S<sub>3</sub>
- FeS<sub>2</sub>
- Fe<sub>3</sub>S<sub>2</sub>

The correct answer is FeS. Iron(II) has a charge of +2 and sulfur has a charge of -2. The charges cancel out, resulting in the formula FeS.

# Cbse Salt Analysis Manual

**JA Banks**



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