

## Calculating pH

How will it be used for the concentration differences here?

Why?

As buffers and other substances control pH in organisms, it is important to understand the ability of buffers to resist change. This property is called **buffering capacity** and is usually expressed as millimoles per litre of buffer that will change the pH of a solution from 7.35 to 7.45 or vice versa. Buffering capacity is related to the concentration of the buffer components. The more buffer (in pH of 7.35 to 7.45), the more buffering capacity. It is important to understand the relationship between pH and buffering capacity in organisms.

### Table 1 – Ion Concentrations for Blood and Urine

Species	Substance	Location (Units in Molar)	Normal	Range	Normal
1	H <sub>2</sub> O	Water	5.5 × 10 <sup>-2</sup> M	5.5 × 10 <sup>-2</sup> M	Normal
2	Cl <sup>-</sup>	Water	1.0 × 10 <sup>-2</sup> M	1.0 × 10 <sup>-2</sup> M	Normal
3	Na <sup>+</sup>	Water	1.35 × 10 <sup>-2</sup> M	1.35 × 10 <sup>-2</sup> M	Normal
4	H <sub>2</sub> O	Water	5.5 × 10 <sup>-2</sup> M	5.5 × 10 <sup>-2</sup> M	Normal
5	Cl <sup>-</sup>	Water	1.0 × 10 <sup>-2</sup> M	1.0 × 10 <sup>-2</sup> M	Normal
6	Na <sup>+</sup>	Water	1.35 × 10 <sup>-2</sup> M	1.35 × 10 <sup>-2</sup> M	Normal
7	H <sub>2</sub> O	Water	5.5 × 10 <sup>-2</sup> M	5.5 × 10 <sup>-2</sup> M	Normal
8	Cl <sup>-</sup>	Water	1.0 × 10 <sup>-2</sup> M	1.0 × 10 <sup>-2</sup> M	Normal
9	Na <sup>+</sup>	Water	1.35 × 10 <sup>-2</sup> M	1.35 × 10 <sup>-2</sup> M	Normal
10	H <sub>2</sub> O	Water	5.5 × 10 <sup>-2</sup> M	5.5 × 10 <sup>-2</sup> M	Normal
11	Cl <sup>-</sup>	Water	1.0 × 10 <sup>-2</sup> M	1.0 × 10 <sup>-2</sup> M	Normal
12	Na <sup>+</sup>	Water	1.35 × 10 <sup>-2</sup> M	1.35 × 10 <sup>-2</sup> M	Normal

1. What are the normal pH values?

2. What are the normal pH values?

3. What are the normal pH values?

4. What are the normal pH values?

5. What are the normal pH values?

6. What are the normal pH values?

7. What are the normal pH values?

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9. What are the normal pH values?

10. What are the normal pH values?

11. What are the normal pH values?

12. What are the normal pH values?

# Calculating Ph Pogil Answer Key

**Padhraic Smyth**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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