

Classroom lecture notes, automotive starting, lighting and ignition, with diagrams

<u>Classroom Lecture Notes Automotive Starting Lighting</u> <u>And Ignition</u>

R. C. Fryer

Classroom Lecture Notes Automotive Starting Lighting And Ignition:

Classroom lecture Notes: Automotive Starting, Lighting and Ignition R. C. Fryer, 1918 **Classroom Lecture** Notes, Automotive Starting, Lighting and Ignition Roy Charles Fryer, 1918 Classroom Lecture Notes, Automotive Starting, Lighting and Ignition; with Diagrams Roy Charles Fryer, 2013-09 This historic book may have numerous typos and missing text Purchasers can usually download a free scanned copy of the original book without typos from the publisher Not indexed Not illustrated 1918 edition Excerpt grows less Tf this discharge continues the gravity of the electrolyte may fall as low as 1 100 or even lower but should never be allowed to fall this low as when gravity reaches this point lead sulphate forms very rapidly on the plates the voltage of the cell becomes very low and the cell in this condition will readily freeze in cold weather The gravity of the cell should never be allowed to drop below 1 250 and should be maintained as nearly as possible at a point between 1 275 and 1 300 The process of charge and the process of discharge should never be carried out at too high a rate as the chemical action taking place in the cell will produce a high degree of temperature and the plates will warp or buckle allowing the active material to fall from them and to accumulate in the sediment chamber This active material would not only mean a loss in the capacity of the plates but would also accumulate until it might come in contact with the plates and form an internal short circuit of the cell and the cell would immediately discharge A storage battery should never be allowed to stand in a discharged condition as the plates will sulphate very rapidly and sulphation is the worst enemy of the lead acid storage battery. The process of mixing acid and water must be carried out very slowly and very carefully. Never add water to acid Always mix water and acid by pouring the acid slowly into the water and stirring continually as in the chemical combination of the two a large amount of heat is evolved and were you to pour the water into the acid you might be severely injured Acid must always be handled in porcelain glass earthenware or lead lined vessels Caution Note this paragraph very carefully CARE OF STORAGE BATTERIES Classroom Lecture Notes, Automotive Starting, Lighting and Ignition Roy Charles Fryer, 1918 **Classroom Lecture Notes** Roy Charles Fryer,1918 **Classroom Lecture Notes**, **Automotive Starting, Lighting and Ignition** Roy Charles Fryer, 2015-08-23 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work was reproduced from the original artifact and remains as true to the original work as possible Therefore you will see the original copyright references library stamps as most of these works have been housed in our most important libraries around the world and other notations in the work This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work As a reproduction of a historical artifact this work may contain missing or blurred pages poor pictures errant marks etc Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping

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