

## CLEAPSS advice on handling liquid nitrogen in schools

(<http://www.cleapss.org.uk/>)

*This is an extract from Section 11 of the CLEAPSS Laboratory Handbook which is supplied to all member secondary schools and colleges (but not to primary schools) on the CLEAPSS Science Publications CD-ROM and on paper.*

- |   |  |   |
|---|--|---|
| m | <b>Liquid nitrogen</b>   | <p>The use of liquid nitrogen allows exciting and thought-provoking demonstrations but there are significant hazards, the risks from which must be controlled. They are:</p> <ul style="list-style-type: none"><li>• asphyxiation in oxygen-deficient atmospheres;</li><li>• fire in oxygen-enriched atmospheres;</li><li>• cold burns, frost bite &amp; hypothermia from the intense cold;</li><li>• over-pressurisation from the large volume expansion of the liquid;</li><li>• manual-handling accidents if using large (25 litre) volumes.</li></ul> <p>There are also risks associated with the transport of liquid nitrogen.</p>   |
| m | <b>Asphyxiation in oxygen-deficient atmospheres</b>                      | <p>Small amounts (not exceeding 2 litres) of liquid nitrogen can be safely used in school laboratories, without risk of asphyxiation. It would be prudent to ensure good ventilation. However, if liquid nitrogen is kept in preparation rooms, stores and other confined spaces, these must be ventilated as much as possible. In the event of a spill in a confined space, the room should be evacuated. If liquid nitrogen has been kept in a confined space, when retrieving it, a second person should be present in case of asphyxia.</p>   |
| m | <b>Fire in oxygen-enriched atmospheres</b>                               | <p>The boiling point of oxygen is <math>-183^{\circ}\text{C}</math>, higher than that of nitrogen, <math>-196^{\circ}\text{C}</math>. Liquid nitrogen may condense liquid oxygen from the surrounding air. Substances will burn more fiercely and ignite. Provided that the quantities of liquid nitrogen are small (eg, 2 litres) there is not much risk of significant oxygen enrichment. Never leave a test tube, or any other container surrounded by liquid nitrogen, for long periods.</p>  |
| m | <b>Cold burns, frost bite and hypothermia from the intense cold</b>      | <p>When handling liquid nitrogen, eye protection (preferably goggles or a face shield) must be worn as well as non-absorbent, leather gloves. Shoes and gloves need to be easily removable but open-toed shoes should <b>not</b> be worn. Demonstrators need to be vigilant about the possibility of splashes of liquid nitrogen becoming entrapped in clothing, etc.</p>   |
| m | <b>Over-pressurisation from the large volume expansion of the liquid</b> | <p>2 litres of liquid nitrogen will become 1366 litres of gaseous nitrogen. In an enclosed vessel this will give rise to a huge increase in pressure. Because liquid nitrogen is cold, it will condense any moisture in the atmosphere, possibly forming an ice plug which may seal an open vessel, causing the pressure to build up. To our knowledge this has caused at least two explosions in schools/colleges. Liquid nitrogen must be transported, kept and used only in a vented Dewar flask, specially designed for cryogenic work. Do <b>not</b>, under any circumstances, use ordinary vacuum flasks. (Suitable small Dewars are available for less than £100 (2005 prices), eg, from Camlab.</p>   |
| m | <b>Transport of liquid nitrogen</b>                                      | <p>Under the <i>Transportable Pressure Vessels Regulations</i>, liquid nitrogen must only be transported in vessels which are suitable for this purpose (see above). Open vessels are <b>not</b> suitable, because of the risk of splashing and spills.</p> <p>If liquid nitrogen is transported by road, the <i>Carriage of Dangerous Goods and Use of Transportable Pressure Equipment Regulations</i> require that the driver shall have had special training. The BCGA Code of Practice CP30 <i>The Safe Use of Liquid Nitrogen Dewars up to 50 litres</i> (British Compressed Gases Association, 2000) states that Dewars must be transported separately from driver or passengers. Flat-back pick-ups, trailers or vehicles fitted with a separating bulkhead should be used. These two requirements mean it is unlikely that schools will be able to obtain free supplies from a</p> |

# Cleapss Laboratory Manual

**Ken Gadd**



## **Cleapss Laboratory Manual:**

*CLEAPSS Laboratory Handbook* CLEAPSS School Science Service, **Aspects of Teaching Secondary Science** Sandra Amos, Richard Boohan, 2003-09-02 A key new textbook which is part of a new series co published with The Open University Written to be used in conjunction with its counterpart in the Teaching in the Secondary School series Between them they address both the theoretical and practical issues in science teaching Examples of good practice are underpinned by reference to research and other literature **Advanced GNVQ Science** M. W. Brimicombe, 1995 This is a student resource book covering the eight mandatory units and core skills at Advanced Level Developed in association with the RSA Examinations Board it provides information and techniques to support assignments case studies to illustrate real life science and exemplar assignments **That's Chemistry!** Jan Rees, 2000 That s Chemistry is a concise manual of ideas activities and investigations about the science of materials and their properties for teachers to use with primary age children All experiments in this book have been trialled in schools It is designed for both specialist and non specialist primary teachers to encourage interest and enthusiasm in a new generation of scientists **Teaching Secondary Science** Keith Ross, Liz Lakin, Janet McKechnie, Jim Baker, 2010-02-25 Now fully updated in its third edition Teaching Secondary Science is a comprehensive guide to all aspects of science teaching providing a wealth of information and ideas about different approaches With guidance on how children understand scientific ideas and the implications this has on teaching teachers are encouraged to construct their own meanings and become reflective in their practice Relating science to government agendas such as the National Strategies Assessment for Learning and Every Child Matters this new edition reflects and maps to changes in national standards Key features include illustrative examples for use in the classroom theoretical grounding linked to practical application the pros and cons of different approaches M Level support materials additional section on earth atmosphere and space advice on teaching difficult ideas education for sustainable development managing the science classroom and health and safety support for talk for learning and advice on numeracy in science Presenting an environmentally sustainable global approach to science teaching this book emphasises the need to build on and challenge children s existing ideas so they better understand the world in which they live Essential reading for all students and practising teachers this invaluable book will support those undertaking secondary science PGCEs and provides material suitable for those studying at M Level **CLEAPSS School Science Service Laboratory Handbook** CLEAPSS School Science Service, 1989\* **Salters Higher Chemistry** Gwen Pilling, 1999 This work is the accompanying teacher s book to the student book and gives the answers to all the questions in the student book together with details of how the student book delivers all the content statements in Higher chemistry **The Nature of Science** Dorothy Warren, 2001 There are many different approaches to science Many students in school have a view of science that is far removed from the real world practice of science and they too often consider science to be a defined body of knowledge and scientists to know all the

answers Through a series of exercises this book tries to give students an awareness of the processes of science and of the nature of science as a changing body of knowledge with uncertainties and much remaining to be discovered It is hoped that by using it teachers can communicate to their students some of the wonder and excitement of science and encourage the development of future generations of scientists The Nature of Science includes a range of activities designed to look at different aspects of the nature of science and to teach investigative skills to 11 19 year old students It contains teachers notes background information photocopiable student worksheets and answers Health, Safety and Risk Dorothy Warren,2001 Health Safety and Risk has been produced in order to help teachers bring about a better understanding of these concepts in their students so that they can be properly aware of the risks associated with the world in which they live and how to minimise them **The School Science Review** ,2007 **Applied Science** Ken Gadd,2003-07 A brand new full colour student resource that precisely matches the new GCSE Double Award specifications and encapsulates the distinctive teaching and learning styles of this new qualification Highly accessible text design allows students to dip in and out for information as and when they need it and to progress with ease through the course Climate Change Dorothy Warren,2001 This is a flexible resource and can be used to study both ideas and evidence and the nature of science and also when teaching key skills **Paracetamol** Frank Ellis,2002 Brief Contents How to use this book Background information Paracetamol is a common compound The history of paracetamol Experimental and investigation section The extraction and purification of paracetamol from tablets The preparation of paracetamol The quantitative analysis of various formulations of paracetamol Using thin layer chromatography to investigate paracetamol Teachers notes The toxicity of paracetamol Apparatus lists and answers **Chemists in a Social and Historical Context** Dorothy Warren,2001 This book focuses on strategies for teaching about people in chemistry and is an introduction to some chemists who played a role in the development of major ideas in the subject Midwest **In Search of More Solutions** Janet Taylor,1995 In Search of More Solutions provides material for chemical egg race competitions that help to put the fun back into chemistry and gives students that little bit extra to motivate them to follow a scientific or technical career Aimed at those following post 16 chemistry courses the book encourages students to design their own experiments in relation to the tasks set and as they overcome the obstacles within the tasks their inventiveness and enthusiasm flourishes Fifty activities are described providing students with opportunities to develop their problem solving skills and teachers with ideas for assignments and investigations The tasks set in this book have been trialled in schools and colleges around the UK Science Learning, Science Teaching Jerry Wellington,Gren Ireson,2013-02-28 Now fully updated in its third edition Science Learning Science Teaching offers an accessible practical guide to creative classroom teaching and a comprehensive introduction to contemporary issues in science education Aiming to encourage and assist professionals with the process of reflection in the science classroom the new edition examines the latest research in the field changes to curriculum and the latest standards for initial teacher

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