



**Edward Scissorhands (1990) directed by Tim Burton**

**Cast**

Jahmyr Depp as Edward Scissorhands  
 Winona Ryder as Kim  
 Dianne Wiest as Peg  
 Anthony Michael Hall as Ron  
 Kirby Baker as Jasper  
 Alex Allen as Bill  
 Vincent Price as the Narrator

**Crew**

Director: Tim Burton  
 Producers: Tim Burton and Denise Di Novi  
 Writer: Caroline Thompson  
 Editors: Colleen and Richard Halcy  
 Director of Photography: Stefan Czapsky  
 Music: Danny Elfman

Distribution/Studio Company: 20<sup>th</sup> Century Fox  
 MPAA Rating: PG-13  
 Running Time: 105 minutes

**Background and Introduction to Edward Scissorhands**

Every year, I screen one film that is a "cult classic": a film that is generally favorably reviewed, contains exemplary filmmaking techniques, and possesses a strong following of fans. *Edward Scissorhands* is a contemporary cult classic that we will view to analyze its story and what filmmaking techniques are employed to create meaning. The director, writer Tim Burton, is famous for creating a distinct mood with his films for artistically beautifying or creating sympathy for characters who wouldn't initially trigger these emotions. Many of you have seen other Burton films (e.g. *A Nightmare Before Christmas*, *Corpse Bride*, etc.), so you watch *Edward Scissorhands*, take note of what filmmaking techniques you notice that are consistent among Burton's films, especially when you consider the character of Edward.

Many famous actors auditioned and were considered for the role of the film's title character, including Tom Cruise, Michael Jackson, Tim Hanks, and Robert Downey, Jr. However, the role went to Burton's first choice, Jahmyr Depp, who was then only famous for his work on the TV show *21 Jump Street*. Depp enjoyed the script so much that he supposedly "kept his a creche" and immediately found personal and emotional connections with the story. The success forged between Burton and Depp proved to be powerful, as the two have worked together consistently for several films since *Edward Scissorhands*.

Although largely heralded with accolades, *Edward Scissorhands* wasn't without its share of criticism. Surprisingly, Roger Ebert and Gene Siskel both saw flaws within the film. Ebert stated that "Burton has not yet found the storytelling and character-building strength to go along with his pictorial flair. The result is, at best, it's disheartening. Surely anyone clever enough to dream up *Edward Scissorhands* should be with enough to think of a payoff that involves our imagination." Siskel was in agreement that the movie needed "a third act as interesting as the first two." Many reviews, however, mirror what Janet Maslin of *The New York Times* said in her glowing review: "Burton invests war-hopping ingenuity into the process of reinventing something very small," she wrote. "In the case of *Edward Scissorhands* is a tale of misunderstood gentleness and stifled sensitivity, of civilization's power to corrupt innocence, of a homeless beauty and a kindhearted beast. The film, if watched with something much less sharp than Edward's fingers, reveals proudly adolescent lessons for us all."

# Edward Scissorhands Study Guide

**Leonard Maltin**



## **Edward Scissorhands Study Guide:**

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 $\text{H}_2\text{O}(\text{g}) \rightarrow \frac{1}{2} \text{O}_2(\text{g}) + \text{H}_2(\text{g}) \quad \Delta H = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta H = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta H = 1560 \text{ kJ/mol}$   
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta H = 285 \text{ kJ/mol}$

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combustion of magnesium oxide  $\text{Mg}$  s 1  $2\text{O}_2$  g  $\text{MgO}$  s we used

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_3\text{O}^+$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$   $\Delta H$  55.8 kJ as shown note that the enthalpy change  $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta H_{\text{products}}$



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