

## TYPES OF CHEMICAL REACTIONS LAB PART#1

**I. Purpose:** To view the actual chemical reactions, write the correct balanced chemical equation, and type of chemical reaction.

**Connection to Business:** Businesses utilize many types of chemical reactions to produce, maintain, and sterilize their product or to prevent pollution.

**II. Materials:** burner, test tube, metal tongs (only....no rubber ends), microspatula, evaporating dish, wood splints

**Chemicals:** copper (wire), magnesium (ribbon), copper (II) carbonate

**SAFETY GOGGLES!**

### III. Procedure: Part #1

1. Obtain materials.
2. Use the fine sandpaper to clean a piece of copper wire until shiny.
3. Set up a Bunsen burner.
4. Using tongs, hold the wire in the hottest part of the burner flame for 1-2 minutes. Record your observation of the wire's appearance afterwards.
5. Place an evaporating dish near the Bunsen burner and repeat step #4 except place the remaining magnesium into the evaporating dish.  
(WARNING: AVOID LOOKING DIRECTLY AT THE FLAME)
6. Place 2 heaping microspatulas of copper (II) carbonate into a clean, dry test tube.
7. Using a test tube holder, hold the test tube (at a 45° angle pointed away from people) over the Bunsen burner flame for 3 minutes or until completely black.  
(Allow this to cool slightly before placing into the wood test tube holder. )
8. Light a wood splint and place inside the test tube. Record your observations.
9. Dispose of all the metals and burnt powders in a zip-lock bag and place in the trash.
10. Clean and return all lab equipment.

### IV. Data/Questions: Part #1 Data Table

Sample	Appearance Before Reaction	Appearance After Reaction
1. Copper	<i>Answer: shiny, reddish brown</i>	<i>Answer: black, powder coating</i>
2. Magnesium	<i>Answer: silver, tinsel</i>	<i>Answer: black ashes, bright light</i>
3. Mercury (II) Oxide (will not perform experiment)	<i>Reddish powder</i>	<i>Liquid mercury formed; flame burned brighter when placed in the test tube</i>
4. Copper (II) Carbonate	<i>Answer: green powder</i>	<i>Answer: decomposes into gas CO<sub>2</sub> and black residue</i>

# Describing Chemical Reactions Answer Key Guided Practice

**RJ Alexander**



## **Describing Chemical Reactions Answer Key Guided Practice:**

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises *Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice* Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 Barron's Science 360 Chemistry is your complete go to guide for everything chemistry This comprehensive guide is an essential resource for High school and college courses Homeschooling Virtual Learning Learning pods Inside you'll find Comprehensive Content Review Begin your study with the basic building block of chemistry and build as you go Topics include atomic structure chemical formulas electrochemistry the basics of organic chemistry and much more Effective Organization Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs Clear Examples and Illustrations Easy to follow explanations hundreds of helpful illustrations and numerous step by step examples make this book ideal for self study and rapid learning Practice Exercises Each chapter ends with practice exercises designed to reinforce and extend key skills and concepts These checkup exercises along with the answers and solutions will help you assess your understanding and monitor your progress Access to Online Practice Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come

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**Essential Practices for Managing Chemical Reactivity Hazards** Robert W. Johnson, Steven W. Rudy, Stephen D. Unwin, 2010-08-13 In its recent investigation of chemical reactivity accidents the US Chemical Safety Board noted a gap in technical guidance and regulatory coverage This volume closes the gap in technical guidance helping small and large companies alike identify address and manage chemical reactivity hazards It guides the reader through an analysis of the potential for chemical reactivity accidents to help prevent fires explosions toxic chemical releases or chemical spills This volume is applicable to processes at any scale and is particularly useful for chemists safety managers and engineers involved in scale up An enclosed CD ROM provides portable checklists analysis tools and a list of additional references Note CD ROM DVD and other supplementary materials are not included as part of eBook file

**Teacher's Wraparound Edition: Two Biology Everyday Experience** Albert Kaskel, 1994-04-19

**Holt Chemistry** Ralph Thomas Myers, 2004

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**Matter, Building Block of the Universe**, 1993

*Platers' Guide*, 1919

Matter Prentice-Hall Staff, 1994

*Study Guide to Accompany Calculus for the Management, Life, and Social Sciences* Clyde Metz, 1984-01-01

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**Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in

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Mohd. Zafar, Moaz Siddiqui, Rachna Rani, Reetika Gulati, Sonal Chauhan, Maukta Gigras, 2021-11-25

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Matter and Laws of Chemical Combinations Chemical Equations and Stoichiometry States of Matter Gaseous and Liquid States States of Matter Solid State Atomic Structure Radioactivity and Nuclear chemistry Chemical Bonding and Molecular Structure Chemical Thermodynamics Solutions Chemical Equilibrium Ionic Equilibrium Redox Reactions Electrochemistry Chemical Kinetics Adsorption Colloidal State Periodic Classification and Periodic Properties Principles and Process of Metallurgy

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