

# Dr. Lee Salk's Super Sitters



A complete home system for you and your babysitter

VHS



Better Care for Your Children and  
Greater Peace of Mind for You!

- Super Sitters promotes your child's safety, well-being and self-esteem.
- Dr. Salk teaches sitters to "take charge" with gentle, non-threatening behavior.
- Sitters learn to provide the best care at all times — not just in emergencies.

# Dr Lee Salks Super Sitters Vhs

**Lee Salk, Jay Litvin**



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*Be a Super Sitter* Lee Salk, Jay Litvin, 1998 Gives advice to young babysitters on getting jobs keeping a business going and safe handling of babies and children      **How to be a Super Sitter** Jay Litvin, Lee Salk, 1990 Gives advice to young babysitters on getting jobs keeping a business going and safe handling of babies and children      **Bowker's Complete Video Directory** ,2000      **Bowker's Directory of Videocassettes for Children 1999** R R Bowker Publishing, Bowker, 1999-03      **The Video Source Book** ,1992      **Glamour** ,1991      **The Elementary School Library Collection** Lauren K. Lee, Linda L. Homa, 1994 Cited in Sheehy and Walford this comprehensive reference recommends print and audiovisual materials as well as microcomputer software and CD ROM products for preschool through sixth grade children The present edition includes 12 294 recommended titles 3 070 being new listings Each entry notes the format s available and provides cataloging and ordering information a critically descriptive annotation interest and reading level estimates and priority for acquisition For school public and academic librarians and preservice and inservice faculty Published by the Brodart Company 500 Arch St Williamsport PA 17705 Annotation copyright by Book News Inc Portland OR  
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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{Co} \rightarrow \text{Fe} + \text{Co}^{2+}$   
 oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni} \rightarrow 3 \text{ Ag} + \text{Ni}^{2+}$  oxidation reduction  $25 \text{ Cu} + \text{Pb} \rightarrow \text{Cu}^{2+} + \text{Pb}$  oxidation reduction  $26 \text{ O}_2$

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 $5\text{CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$   $5\text{CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$

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