# Quiz & Worksheet - Double Displacement Reaction

 What are the correct products for the following reaction? (also include the correct coefficient for a balanced reaction)

- AgSO<sub>4</sub> + NaNO<sub>3</sub>
- Ag<sub>2</sub>SO<sub>4</sub>+ 2NaNO<sub>3</sub>
- Ag<sub>2</sub>SO<sub>4</sub>+ NaNO<sub>3</sub>
- Ag<sub>2</sub>SO<sub>4</sub> + Na<sub>2</sub>NO<sub>3</sub>

2. What are the ions and the correct charges for the following compound:

## PbCl<sub>2</sub>

- Pb2+ Cl-
- Pb+ Cl<sup>2</sup>·
- Pb<sup>2</sup>- Cl<sup>2</sup>-
- Pb<sup>2+</sup> Cl<sup>2-</sup>

3. What type of reaction is the following?

$$PbCl_2(aq) + K_2SO_4(aq) \longrightarrow 2KCl(aq) + PbSO_4(s)$$

- neutralization
- precipitation
- gas formation
- combustion.

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