Class	General Formula	Example	Common Name (Systematic Name)	Common Suffix/Prefix (Systematic)
		Hydrocarbons		
Alkanes	RH	CH ₃ CH ₃	ethane	-ane
Alkenes	RR'C-CR"R"	H_2C — CH_2	ethylene (ethene)	-ene
Alkynes	RC=CR'	нс—сн	acetylene (ethyne)	(-yne)
Arenes	ArH°		benzene	-ene
	Hale	ogen-Containing Comp	ounds	
Alkyl halides	RX	CH ₃ CH ₂ CI	ethyl chloride (chloroethane)	halide (halo-)
Aryl halides	ArX"		chlorobenzene	halo-
	Oxy	gen-Containing Compo	ounds	
Alcohols	ROH°	CH ₃ CH ₂ OH	ethyl alcohol (ethanol)	-ol
Phenois	ArOH*	>он	phenol	-ol
Ethers	ROR"	H ₃ CH ₂ COCH ₂ CH ₃	diethyl ether	ether
Aldehydes	RCHO	CH ₃ CH	acetaldehyde (ethanal)	-aldehyde (-al)
Ketones	RR'C=O	CH ₃ CCH ₃	acetone (2-propanone)	-one
Carboxylic acids	RCO₂H	CH ₃ COH	acetic acid (ethanoic acid)	-ic acid (-oic acid)
		arboxylic Acid Derivati	ves	
Esters	RCO₂R*	CH ₃ COCH ₃	methyl acetate (methyl ethanoate)	-ate (-oate)
Amides	RCONHR'	CH ₃ CNHCH ₃	N-methylacetamide	-amide
	Nitr	ogen-Containing Comp	ounds	
Amines	RNH ₂ , RNHR", RNR'R"	CH ₃ CH ₂ NH ₂	ethylamine	-amine
Nitriles	RC≡N	H ₃ CC≡ N	acetonitrile	-nitrile
Nitro compounds	ArNO2"	NO ₂	nitrobenzene	nitro-
R indicates an alkyl gro	up ^b Ar indicates an <i>aryl</i>	group.		

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G Thomas



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