

Chemistry EOC Review – Answer Key

- 1) a. 68,200,000 cg b. 5000 mm c. 0.0125 kL d. 0.548 dm
 - 2) ~21.5 mL
 - 3) production of heat & light, formation of a gas, formation of a precipitate, color change
 - 4) $33.64 \text{ g} / 4.28 \text{ ml} = 7.86 \text{ g/ml} = \text{Fe}$
 - 5) m.p. $-183^{\circ}\text{C} = \text{CH}_4$ (methane)
 - 6) H_2 – splint test (burning splint ignites hydrogen)
 O_2 – splint test (oxygen reignites glowing splint)
 H_2O – cobalt chloride paper (paper changes from blue to pink in the presence of water vapor)
 CO_2 – splint test (carbon dioxide extinguishes burning splint);
 limewater test (limewater turns cloudy when carbon dioxide is bubbled through it,
 producing CaCO_3)
 - 7) The protons are positively charged particles found inside the nucleus along with the neutrons, which have no charge. The total number of these nucleons in an atom is the mass # / atomic mass of the atom. The electrons are the negatively charged particles found orbiting in "clouds" around the nucleus. Electrons can be gained by the atom to form anions or lost to form cations.
 - 8) The thing that makes any element on the periodic table different from all other elements is the number of protons in its nucleus. Isotopes are atoms of the same type of element which have different masses because they have different numbers of neutrons.
 - 9) a. $^{39}_{19}\text{K}^{+}$
 b. $^{57}_{26}\text{Fe}$
 c. $^{33}_{16}\text{S}^{2-}$
- 100

<u>chemical symbol</u>	<u>atomic number</u>	<u>mass number</u>	<u># of protons</u>	<u># of electrons</u>	<u># of neutrons</u>	<u>net charge</u>
K ⁺	19	42	19	18	23	+1
Sr	38	88	38	38	50	0
F ⁻	9	20	9	10	11	-1
P	15	31	15	15	16	0
Ca ²⁺	20	42	20	18	22	+2

- 11) uranium - 238 $238 - 92 = 146$ neutrons
- 12) (1) 199.59g
(2) They contain the same # of protons (they are the same element), but different #s of neutrons (electrons don't matter, b/c they have virtually no mass), & thus different masses

Eoc Chemistry Practice 7b Answers

Terry C. Jones



Eoc Chemistry Practice 7b Answers:

Roadmap to the Virginia SOL Princeton Review, 2005 *Roadmap to the Virginia SOL EOC Chemistry* includes strategies that are proven to enhance student performance The experts at The Princeton Review provide content review of the crucial material most likely to appear on the test detailed lessons complete with test taking techniques for improving test scores 2 complete practice Virginia SOL EOC Chemistry tests

Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-01-19 *Organic Chemistry* 13th Edition continues Solomons Fryle and Snyder s tradition of excellence in teaching and preparing students for success in both the classroom and beyond Central to the authors is their approach in emphasizing organic chemistry s relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

Florida Biology 1 End-of-Course Assessment Book + Online John Allen, 2013-03-26 *Taking the Florida Biology 1 End of Course Exam* Then You Need REA s *Florida Biology 1 End of Course Test Prep with Online Practice Exams* If you re facing the Florida Biology 1 End of Course exam and are concerned about your score don t worry REA s test prep will help you sharpen your skills and pass this high stakes exam REA s *Florida Biology 1 End of Course test prep* provides all the up to date instruction and practice you need to improve your skills The comprehensive review features easy to follow examples that reinforce the concepts tested on the Biology 1 End of Course exam Our test prep is ideal for classroom group or individual study Tutorials and targeted drills increase your comprehension Color icons and graphics throughout the book highlight important concepts and tasks REA s test taking tips and strategies give you the confidence you need on test day so you can pass the exam and graduate The book contains two full length practice exams that let you test your knowledge while reinforcing what you ve learned The same two practice tests are also available online at REA s Study Center The online tests give you the additional benefits of instant scoring timed testing conditions and diagnostic score reports that pinpoint your strengths and weaknesses Each practice test comes complete with detailed explanations of answers so you can focus on areas where you need extra review This book is a must for any Florida student preparing for the Biology 1 End of Course exam About the Exam The Florida Biology I End of Course exam measures middle and high school student achievement of the Next Generation Sunshine State Standards All public school students are required to pass the exam in order to receive a high school diploma

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