Density Worksheet Physical Science D=m/V

Substance	Density (g/cm²)	Substance	Density (g/essa)
Oxygen	0.00933	Aluminam	2.70
Hydrogen	0.000084	Iron	7.87
Ethanol	0.785	Copper	8.96
Benzene	0.880	Silver	10.5
Water	1.000	Lead	31-34
Magnesium	3.74	Mercury	13.6
Salt (sodium chloride)	2.16	Gold	19.32

 The ratio of an object's mass to its ______ is called the devisity of the object.

 A kilogram of lead occupies a much smaller volume than a kilogram of water, because has a much higher density.

 For the masses and volumes indicated, calculate the density in grams per cubic centimeters.

a. mass = 453 g; volume = 225 cm3

b. mass = 5.0 g; volume = 10.0 cm3

c. mass = 26.1 g; volume = 2.0 mL

4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.

A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

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M Walker

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Embracing the Beat of Term: An Emotional Symphony within **Density Worksheet Physical Science If8767**

In a world used by monitors and the ceaseless chatter of immediate transmission, the melodic elegance and psychological symphony developed by the written term frequently disappear into the backdrop, eclipsed by the constant sound and distractions that permeate our lives. However, located within the pages of **Density Worksheet Physical Science If8767** an enchanting fictional value full of raw feelings, lies an immersive symphony waiting to be embraced. Crafted by an elegant composer of language, that charming masterpiece conducts viewers on a psychological journey, well unraveling the hidden songs and profound influence resonating within each cautiously constructed phrase. Within the depths of this poignant assessment, we will explore the book is central harmonies, analyze their enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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