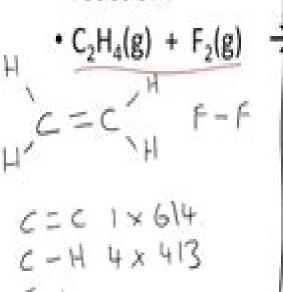
Using bond energy data

Use bond energies to determine the energy change for the following

reaction:



 $C_2H_4F_2(g)$

| Bond | Bond energy (kj/mol) |
|------|----------------------|
| C=C | 614 |
| C-C | 348 |
| C-H | 413 |
| C-F | 485 |
| F-F | 155 |

Energy change = ENERGY IN - ENERGY OUT

Calculating Bond Energy

Camilla Rothe



Reviewing Calculating Bond Energy: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "Calculating Bond Energy," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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Calculating Bond Energy Introduction

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