

Example 1: Propane is commonly burned in homes for heat in the winter. It combusts according to the following chemical equation:



Determine the molar enthalpy change for the reaction using the standard heats of formation of the chemical involved in the process.

$$\Delta H_{\text{rxn}} = \sum n \Delta H_{\text{f}}^{\circ} \text{products} - \sum n \Delta H_{\text{f}}^{\circ} \text{reactants}$$

\uparrow
 mol

Chemical	$\Delta H_{\text{f}}^{\circ}$
$\text{C}_3\text{H}_8 (\text{g})$	$-393.5 \text{ kJ} \cdot \text{mol}^{-1}$
$\text{O}_2 (\text{g})$	$0 \text{ kJ} \cdot \text{mol}^{-1} \leftarrow$
$\text{CO}_2 (\text{g})$	$-241.8 \text{ kJ} \cdot \text{mol}^{-1} \rightarrow$
$\text{H}_2\text{O} (\text{g})$	$-103.8 \text{ kJ} \cdot \text{mol}^{-1}$

$$= \left[3 \text{ mol} (-241.8 \text{ kJ} \cdot \text{mol}^{-1}) + 4 \text{ mol} (-103.8 \text{ kJ} \cdot \text{mol}^{-1}) \right]$$

$$- \left[1 \text{ mol} (-393.5 \text{ kJ} \cdot \text{mol}^{-1}) \right]$$

$$= -2043 \text{ kJ} \cdot \text{mol}^{-1}_{\text{rxn}}$$

Calculating Enthalpy Change Study Guide

Arturo Cuomo



Calculating Enthalpy Change Study Guide:

Calculating Enthalpy Change Study Guide Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Calculating Enthalpy Change Study Guide**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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