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web the balanced chemical equation for this reaction is  $\text{Mg}(\text{s}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + \text{H}_2(\text{g})$  in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

$\text{HNO}_3 + \text{C}_2\text{H}_3\text{O}_2 \rightarrow \text{HC}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$  chemical equation - Aug 11 2023

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