

Quiz & Worksheet - Combined Gas Law

1. Which of the following statements correctly defines the combined gas law?

- ☐ It gives the relationship between pressure and volume of a gaseous substance.
- ☐ It gives the relationship between temperature and volume of a gaseous substance.
- ☐ It gives the relationship between temperature and pressure of a gaseous substance.
- ☐ It gives the relationship of the volume, temperature and pressure of a gaseous substance.

2. Which of the following do you expect to occur to the temperature when a gas that occupies a container suddenly experiences an increase in pressure and volume?

- ☐ The temperature decreases.
- ☐ The temperature remains the same.
- ☐ The temperature increases.
- ☐ The temperature goes below freezing point.

3. According to the combined gas law, which of the following statements is incorrect?

- ☐ Pressure is inversely proportional to volume
- ☐ Pressure is directly proportional to temperature
- ☐ Volume is directly proportional to temperature.
- ☐ Pressure is directly proportional to volume

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