Varme				
Period	Date	1	/	

# 12 • The Gas Laws

#### THE COMBINED GAS LAW

In practical terms, it is often difficult to hold any of the variables constant. When there is a change in pressure, volume and temperature, the combined gas law is used.

$$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$$
 or  $P_1 \times V_1 \times T_2 = P_2 \times V_2 \times T_1$   
 $K = {}^{\circ}C + 273$ 

Serater

Complete the following chart.

	P <sub>1</sub>	$\mathbf{V_1}$	T <sub>1</sub>	P <sub>2</sub>	$V_2$	T <sub>2</sub>
1	1.50 atm	3.00 L	293K	2.50 atm		3051€ -30.0°C
2	720. torr	256. mL	298 K -25.0°€-		250. mL	3 23 K -50.0 ℃
3	600. mmHg	2.50 L	325 K -22.0°€	760. mmHg	1.80 L	
4		750. mL	273℃ -0.00°C	2.00 atm	500. mL	2.48 K _25.0°C
5	95.0 kPa	4.00 L		101. kPa	6.00 L	471. K or 498. °C
6	650. torr	8	373 K -100.℃	900. torr	225. mL	423 K -150. °€
7	850. mmHg	1.50 L	288 K -15.0°€		2.50 L	303 K
8	125, kPa	125. mL		100. kPa	100 mL	348 K

# **Combined Gas Law Chart Answers**

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In a world inundated with screens and the cacophony of instant communication, the profound power and emotional resonance of verbal art usually disappear in to obscurity, eclipsed by the constant onslaught of noise and distractions. However, nestled within the lyrical pages of **Combined Gas Law Chart Answers**, a captivating perform of fictional splendor that impulses with fresh thoughts, lies an remarkable journey waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that interesting opus instructions viewers on a mental odyssey, gently revealing the latent potential and profound impact embedded within the elaborate web of language. Within the heart-wrenching expanse of the evocative examination, we shall embark upon an introspective exploration of the book is key subjects, dissect their captivating publishing type, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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