

12 • The Gas Laws**THE COMBINED GAS LAW**

In practical terms, it is often difficult to hold any of the variables constant. When there is a change in pressure, volume and temperature, the combined gas law is used.

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \text{or} \quad P_1 \times V_1 \times T_2 = P_2 \times V_2 \times T_1$$

$$K = ^\circ C + 273$$

*See
Scratch
paper*

Complete the following chart.

	P_1	V_1	T_1	P_2	V_2	T_2
1	1.50 atm	3.00 L	20.0°C 293 K	2.50 atm		30.0°C 303 K
2	720. torr	256. mL	25.0°C 298 K		250. mL	50.0°C 323 K
3	600. mmHg	2.50 L	22.0°C 325 K	760. mmHg	1.80 L	
4		750. mL	0.00°C 273 K	2.00 atm	500. mL	25.0°C 298 K
5	95.0 kPa	4.00 L		101. kPa	6.00 L	498.°C 471. K or
6	650. torr		100.°C 373 K	900. torr	225. mL	150.°C 423 K
7	850. mmHg	1.50 L	15.0°C 288 K		2.50 L	30.0°C 303 K
8	125. kPa	125. mL		100. kPa	100 mL	75.0°C 348 K

Combined And Ideal Gas Laws Answers

Brendan G. Carr



Combined And Ideal Gas Laws Answers:

Decoding **Combined And Ideal Gas Laws Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Combined And Ideal Gas Laws Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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