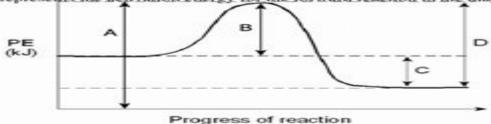
#### PART I Total Value: 50%

# Instructions: Shade the letter of the correct answer on the computer scorable answer sheet provided.

Why does increasing temperature increase reaction rate?

(A) decreases activation energy

- (B) decreases the percentage of effective collisions
- (C) increases activation energy
- (D) increases the percentage of effective collisions
- 2. Which represents the activation energy for the forward reaction in the diagram below?



- (A) A (B) B
  - (C) C (D) D
- 3. Which type of change is best for monitoring the reaction rate of the reaction below?

$$Mg(OH)_2(s) + 2 HCl(aq) \rightarrow 2 H_2O(\ell) + MgCl_2(aq)$$

- (A) colour (B) mass
- (C) pH (D) volume
- 4. Which describes an activated complex?
  - (A) has a lower potential energy than products (B) has a lower potential energy than reactants
- (C) stable species that forms upon collision of reactants
  (D) unstable species that forms upon collision of reactants
- 5. According to the collision theory, why does the reaction below occur in more than one step?

$$C_5H_{12}(g) + 8 O_2(g) \rightarrow 5 CO_2(g) + 6 H_2O(g)$$

- (A) low C₃H₁₂(g) concentration
  - (B) low probability of a nine-particle collision
  - (C) particles collide with insufficient kinetic energy
  - (D) temperature is very high

# **Collision Theory Answer Key**

Scott C. Dulebohn

**Collision Theory Answer Key:** 

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Collision Theory Answer Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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