Substance	Lewis Structure	Molar Mass	Intermolecular Attractive Force
Methanol CH ₃ OH 4+3(1)+6+1+34	H-C-Ö+H	32.042	Hydrogen
Ethanol CH ₃ CH ₂ OH	H-C-C-O-H	46.068	Hydrogen
n-Butanol CH ₃ CH ₂ CH ₂ CH ₂ OH	H-0-0-1 H-0-0-1 H-0-0-1	74.122	Hydrogen
Glycerin C ₃ H _s (OH) ₃	H-C-C-C-H 19-HH 19-H	90.093	Hydrogen
Water H ₂ O	H-0-H	18.015	Hydrogen Bonds

Exploring Intermolecular Forces Lab Answers

Joacim Rocklöv

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