

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

- 1 Write each of the changes listed below in the correct column in the table to show whether they are physical changes or chemical reactions.

Changes: boiling burning melting neutralising

Physical changes	Chemical reactions

- 2 Some of the statements below are true and some are false.

Put a tick (✓) in the box next to the statements that are true.

- ☐ New substances are made during a chemical reaction.
- ☐ New substances are made during a physical change.
- ☐ A chemical reaction is usually easier to reverse than a physical change.
- ☐ A physical change is usually easier to reverse than a chemical reaction.
- ☐ You cannot reverse physical changes or chemical reactions.

- 3 Each observation below is describing a physical change or a chemical reaction.

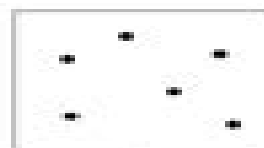
Put a tick (✓) in the correct column to show whether each is a physical change or a chemical reaction.

Observation	Physical change	Chemical reaction
A liquid was set on fire and it burnt with a yellow flame.		
A liquid was cooled down and it changed into a solid.		

- 4 The boxes A and B show gas particles in containers.



A



B

Which box shows the gas at the higher pressure? \_\_\_\_\_

Explain your answer. \_\_\_\_\_

**I can...**

- identify and explain the differences between physical changes and chemical reactions
- use particle theory to explain gas pressure and how it can be changed.

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