The Crucible Act 2 Study Guide Questions and Answers 100% Pass

At the beginning of Act 2, John Proctor says, "It is winter in here yet". Why is this pertinent to what is going on?

John and Elizabeth's relationship is still strained from his affair with Abigail. Elizabeth's demeanor is "frosty" and "cold" towards John.

Why has Mary Warren disobeyed her employers and gone to Salem?

She has been named an official of the court. This makes her feel more important than a servant.

What did Abigail Williams reveal to John Proctor?

Elizabeth reminds him of this; that the girls were just playing around in the woods and no witchcraft was involved.

Why hasn't John Proctor told the court what he knows? What does Elizabeth attribute his not telling?

John worries how he would prove what Abigail said(since they were alone). Also, he is aware of the power she has gained. "If the girl's a saint now, I think it not easy to prove she's a fraud."

What lie did John Proctor tell to Elizabeth which makes her more suspicious of him?

√√He

didn't tell Elizabeth that he was alone with Abigail when they talked.

Crucible Study Guide Questions And Answers

RD Boyd

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town of Salem is in fact engulfed in witchcraft as one by one the girls accuse other townspeople of communing with the devil A trial ensues causing those accused to either deny these allegations or confess thus accusing someone else This cycle finally culminates in the death of several innocent townsfolk The Crucible is a historical dramatization of true events that show reputation is more important than admitting ignorance Class 11-12 Chemistry MCQ (Multiple Choice Questions) Arshad Igbal, 2019-05-17 The Class 11 12 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF College Chemistry MCQ PDF Download Quiz Questions Chapter 1 6 Practice Tests with Answer Key 11th 12th Grade Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 11 12 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry MCO PDF book helps to practice test questions from exam prep notes The Class 11 12 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 11 12 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved guiz guestions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Class 11 12 Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book Grade 11 12 Chemistry MCQs Chapter 1 6 PDF includes college question papers to review practice tests for exams Class 11 12 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Mock Tests Chapter 1 6 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure MCQ Chapter 2 Basic Chemistry MCQ Chapter 3 Chemical Bonding MCQ Chapter 4 Experimental Techniques MCQ Chapter 5 Gases MCQ Chapter 6 Liquids and Solids MCQ The Atomic Structure MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic configuration of elements energy of revolving electron fundamental particles Heisenberg's uncertainty principle hydrogen spectrum magnetic quantum number mass of electron metallic crystals properties Moseley law neutron properties orbital concept photons wave number Planck's quantum theory properties of cathode rays properties of positive rays quantum numbers quantum theory Rutherford model of atom shapes of orbitals spin quantum number what is spectrum x rays and atomic number The Basic Chemistry MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Basic chemistry atomic mass atoms molecules Avogadro s law combustion analysis empirical formula isotopes mass spectrometer molar volume molecular ions moles positive and negative ions relative abundance spectrometer and stoichiometry The Chemical

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