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RJ Shavelson



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$$\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_2 + 4\text{OH}^-$$

$$\text{MnO}_4^- + 2\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + \text{H}_2\text{O}$$

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