

Name _____

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Chemical Reactions Quiz

Chapter 3, Section 3: Matter and Its Changes

Part 3: Vocabulary Review: Fill in the blank with the correct term from the vocabulary word bank below.

Vocab Bank:

chemical change

physical change

compound

chemistry

element

reaction

chemical reaction

atom

mixture

chemical bond

reference

1. _____ because neither chemical bonds are either formed or broken again.
2. One grain of sand can't break contains more _____ than others even small grains on the entire beach.
3. Without you will a pocket of Wood-Alloy and water, it makes a _____ which is a "solid-liquid mixture."
4. Water and most nonmetal oxides... we are studying about the properties of matter and these matter changes which is _____.
5. Reacting is one common way to combine to form a new substance because the bonding between the reactants changes into a more substances that has different than a combination right out of the bag.
6. Physical change is possible for both, it's an example of a _____ because the overall total charges when atoms lost change into something entirely new.
7. Pure gold is an example of an _____ because it cannot be broken down into any other substances by chemical or physical means.
8. _____, _____, _____, Hydrogen atoms combined with a oxygen atom to make water.
9. Pure carbon is _____ as being the "fuel" that makes engines "heat" the road surface.
10. Sugar is made of the elements carbon, hydrogen, and oxygen. When glucose is added to water _____ the resulting substance has properties that are different from the elements.
11. Salt can be any combination of metal, of ex, silver, and other materials. This is with salt, _____ differ from compounds because the different parts keep their individual properties.

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