

Exam:

Name: \_\_\_\_\_

**MULTIPLE CHOICE** Choose the one alternative that best completes the statement or answers the question.

- 1) The total concentration of ions in a 0.250 M solution of HCl is \_\_\_\_\_ 11 \_\_\_\_\_  
A) 0.250 M  
B) 0.500 M  
C) 0.125 M  
D) 0.750 M  
E) essentially zero.
- 2) A weak electrolyte exists predominantly as \_\_\_\_\_ in solution. 21 \_\_\_\_\_  
A) electrons  
B) molecules  
C) atoms  
D) ions  
E) an isotope
- 3) Which of the following are strong electrolytes? 31 \_\_\_\_\_  
HCl  
HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>  
NH<sub>3</sub>  
KCl  
A) HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, NH<sub>3</sub>, KCl  
B) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, KCl  
C) HCl, NH<sub>3</sub>, KCl  
D) HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, KCl  
E) HCl, KCl
- 4) Which of the following are weak electrolytes? 41 \_\_\_\_\_  
HCl  
HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>  
NH<sub>3</sub>  
KCl  
A) HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, NH<sub>3</sub>, KCl  
B) HCl, KCl  
C) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, KCl  
D) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, NH<sub>3</sub>  
E) HCl, HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>, KCl
- 5) When aqueous solutions of \_\_\_\_\_ are mixed, a precipitate forms. 51 \_\_\_\_\_  
A) NaI and KBr  
B) Li<sub>2</sub>CO<sub>3</sub> and CaI  
C) K<sub>2</sub>SO<sub>4</sub> and CaCl<sub>2</sub>  
D) KOH and Ba(NO<sub>3</sub>)<sub>2</sub>  
E) NaBr and AgNO<sub>3</sub>

# Ch 25 Nuclear Chemistry Workbook Answers

**Gary Long**



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