

Science 10  
Balancing Chemical Equations  
Work Sheet

Date Answers

Balance the following equations. Use only whole numbers.

- 1) 2 Na + 2 H<sub>2</sub>O → 2 NaOH +    H<sub>2</sub>
- 2)    Cu + 2 AgNO<sub>3</sub> →    Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag
- 3)    Ca(NO<sub>3</sub>)<sub>2</sub> + 2 NaOH →    Ca(OH)<sub>2</sub> + 2 NaNO<sub>3</sub>
- 4) 2 SO<sub>2</sub> +    O<sub>2</sub> → 2 SO<sub>3</sub>
- 5)    CH<sub>4</sub> + 2 O<sub>2</sub> →    CO<sub>2</sub> + 2 H<sub>2</sub>O
- 6)    AlCl<sub>3</sub> + 3 K → 3 KCl +    Al
- 7) 4 NH<sub>3</sub> + 3 O<sub>2</sub> → 2 N<sub>2</sub> + 6 H<sub>2</sub>O
- 8) 2 CuO → 2 Cu +    O<sub>2</sub>
- 9) 2 HCl +    Na<sub>2</sub>CO<sub>3</sub> → 2 NaCl +    H<sub>2</sub>O +    CO<sub>2</sub>
- 10)    6 CO<sub>2</sub> + 6 H<sub>2</sub>O →    C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> + 6 O<sub>2</sub>

Determine whether the following equations are balanced. If not, give the correct equation.

- |   |          |            |           |
|---|----------|------------|-----------|
| 1) <u>P + 5Cl<sub>2</sub> → PCl<sub>5</sub></u><br><u>2P + 5Cl<sub>2</sub> → 2PCl<sub>5</sub></u>                         | Correct? | Yes        | <u>No</u> |
| 2) <u>H<sub>2</sub>SO<sub>4</sub> + FeS → FeSO<sub>4</sub> + H<sub>2</sub>S</u>   | Correct? | <u>Yes</u> | No        |
| 3) <u>Zn + 2HCl → ZnCl<sub>2</sub> + 2H<sub>2</sub></u><br><u>Zn + 2HCl → ZnCl<sub>2</sub> + H<sub>2</sub></u>            | Correct? | Yes        | <u>No</u> |
| 4) <u>2Fe + O<sub>2</sub> → Fe<sub>2</sub>O<sub>3</sub></u><br><u>4Fe + 3O<sub>2</sub> → 2Fe<sub>2</sub>O<sub>3</sub></u> | Correct? | Yes        | <u>No</u> |
| 5) <u>P<sub>4</sub> + 5O<sub>2</sub> → P<sub>4</sub>O<sub>10</sub></u>  | Correct? | <u>Yes</u> | No        |

# Chemistry Stoichiometry Guide Answers

**WJ Hussar**



## Chemistry Stoichiometry Guide Answers:

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises *Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter [The Complete Idiot's Guide to Chemistry, 3rd Edition](#) Ian Guch, 2011-12-06 This book follows a standard math based chemistry curriculum Author is an award winning teacher who has taught at both the high school and college levels [Chemistry for Nerds Guide Book: Chemistry, Science, Nerd, Geek, Textbook, Guidebook, Study Guide, Educational, STEM, Science Gift](#) Matt Kingsley, Calling all curious minds and science enthusiasts Are you fascinated by the invisible forces that shape our world Do you crave a deeper understanding of the elements molecules and reactions that make up everything around us Then look no further than Chemistry for Nerds Unleash Your Inner Mad Scientist This isn't your typical boring textbook This is a thrilling adventure through the captivating world of chemistry written in a way that's engaging accessible and downright fun Inside these pages you'll discover The secrets of matter From atoms and molecules to the states of matter and the laws that govern them The magic of reactions Explore the explosive world of chemical reactions from kinetics and equilibrium to acids bases and buffers The wonders of the elements Unravel the mysteries of the periodic table and the trends that connect its diverse inhabitants The power of chemistry in action See how chemistry shapes our environment fuels our technologies and even sustains life itself Chemistry for Nerds is packed with Crystal clear explanations Complex concepts are broken down into bite sized pieces making even the most challenging topics easy to grasp Engaging examples and analogies Relate chemistry to everyday life with fun and memorable examples Expert practical tips Put your knowledge into action with helpful tips and tricks for mastering chemistry concepts Whether you're a student a hobbyist or simply curious about the world around you Chemistry for Nerds will ignite your passion for science and unleash your inner mad scientist Get your copy today and start exploring

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