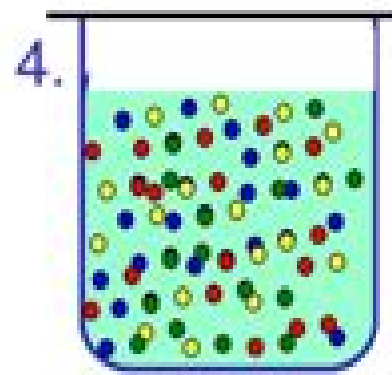
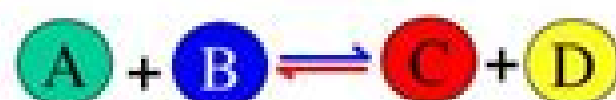
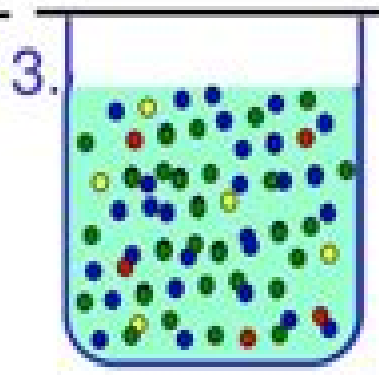
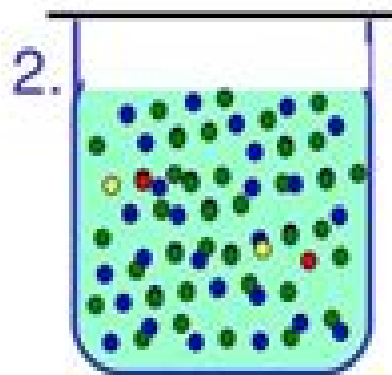
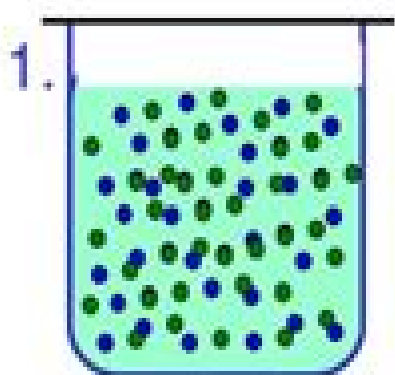


Chemical Equilibrium



1. Reaction begins.
 - No products yet formed.
 - High rate of collisions between A & B.
 - Rate of forward reaction HIGH.

- 2 & 3 Products formed
 - Collisions between reactants decrease.
 - Rate of forward reaction **DECREASES**
 - **Reverse reaction begins.**

4. Rate of forward reaction **EQUAL** to **rate of reverse reaction**.
 - Dynamic equilibrium established.
 - Concentrations constant.

Chemistry Solutions Manual Chemical Equilibrium

Terry C. Jones



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