

Summary 2nd Semester Final Exam Study Guide

Introduction regarding the 2nd Semester Final Exam

Structure of Exam: Multiple choice

Number of questions: 40 items, 100

What is Exam: Check the Final Exam Schedule webpage posted on page 16 for the exam date

Topics to be covered by the 2nd Semester Final Exam

Chemical Reactions involving acids

Measuring Equilibrium

Writing Equilibrium Equations: Homogeneous Equilibrium

Equilibrium Constant Formulas:

$K_{\text{equilibrium}}$

$K_{\text{acid/base}}$

Weak Equilibrium

K_{acid} Equilibrium

K_{base}

$K_{\text{precipitation}}$

Acids/Bases

What is an acid: acids are positive acids in water

Acid types: monoprotic/diprotic

Measurement of the Acid-Base Equilibrium Constant

Acid-Base Equilibrium

Equilibrium and Acid-Base Equilibrium

in Water

Weak Acids and Bases

Equilibrium Equations that involve Acids/Bases and Equilibrium

Weak and Strong Acids/Bases (Strong Acids/Bases)

Equilibrium (in water)

Acids/Bases (in water)

Acids

Properties of Acids

Acid Types

Acids

Properties

Equilibrium Equations

Equilibrium Equations

Chemistry Second Semester Final Exam Study Guide

Thomas Griffiths



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