

**FIRST TEST AUGUST 2013**

CLASS : 11 P.T.  
TIME : 1 Hr. 30 Min.

INSTRUCTIONS :

1. The question paper has 4 parts, A, B, C, and D. All parts are compulsory.  
2. Write balanced chemical equations and draw labelled diagrams wherever required.  
3. Use log tables and simple calculators if necessary. 1% of scientific calculation is not allowed.

**PART - A**

1. Select the correct option from the given choices.

1) Mathematical expression of Henry's law is

$p = k \cdot C$        $p = k \cdot C$        $p = k \cdot C$        $p = k \cdot C$

[illegible]

# Exam

Answer any TWO of the following questions. Each question carries 2 marks. (2x2=4)

Write any two difference between ideal and non-ideal solutions.

The magnetic moment of  $\text{Ni}^{2+}$  ion. ( $g = 2$ )

Write reaction with an example.

Write any two differences between ionic and covalent bonds.

(a) Nitrophosphorus is more acidic than Phosphoric acid. (1+1)

(b)  $\text{PF}_5$  is more acidic than  $\text{PCl}_5$ . (1+1)

# Chemistry Quarterly Exam First Answers

**Michael Rajnik**



## **Chemistry Quarterly Exam First Answers:**

## **Chemistry Quarterly Exam First Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Quarterly Exam First Answers**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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