CHAPTER 17 STUDY GUIDE

Chemical Equilibrium

Section 17.1 Equilibrium: A State of Dynamic Balance

In your textbook, read about chemical equilibrium.

Complete each statement.

- When a reaction results in almost complete conversion of reactants to products, chemists say the reaction goes to
- 2. A reaction that can occur in both the forward and the reverse directions is called a(n)

 is a state in which the forward and reverse reactions balance each other because they take place at equal rates.

- At equilibrium, the concentrations of reactants and products are _______ but that does not mean that the amounts or concentrations are ______.
- 5. Equilibrium is a state of ______, not one of ______

In your textbook, read about equilibrium expressions and constants.

For each statement below, write true or false.

- The law of chemical equilibrium states that at a given pressure, a chemical system may reach a state in which a particular ratio of reactant to product concentrations has a constant value.
 - 7. The equation $H_2(g) + I_2(g) \leftrightarrow 2HI(g)$ is an example of a homogeneous equilibrium.
- If an equilibrium constant has a value less than one, the reactants are favored at equilibrium.
- 9. The value for K_{eq} is constant only at a specific volume.
- If the equilibrium constant for a reaction at 300 K is 49.7, the concentration of the reactants will be greater than the concentration of the products.
- A heterogeneous equilibrium means that reactants and products are present in more than one state.
 - 12. The product of the forward chemical reaction is HI, for the equilibrium expression:

$$K_{eq} = \frac{[HI]^2}{[H_2][I_2]}$$

Chemistry Matter And Change Chapter 12 Study Guide Answers

Judah P. Makonye, Nokulunga S. Ndlovu

Chemistry Matter And Change Chapter 12 Study Guide Answers:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change Chemistry Questions and Answers PDF Arshad Iqbal, The A Level Chemistry Quiz Questions and Answers PDF IGCSE GCE Chemistry Competitive Exam Questions Chapter 1 28 Practice Tests Class 11 12 Chemistry Textbook Questions for Beginners includes revision guide for problem solving with hundreds of solved questions A Level Chemistry Questions and Answers PDF book covers basic concepts analytical and practical assessment tests A Level Chemistry Quiz PDF book helps to practice test questions from exam prep notes The A Level Chemistry Quiz Questions and Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved tests A Level Chemistry Questions and Answers PDF Free download chapter 1 a book covers solved common questions and answers on chapters Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements tests for college and university revision guide Chemistry Interview Questions and Answers PDF Download free eBook's sample covers beginner's solved questions textbook's study notes to practice online tests The IGCSE GCE Chemistry Interview Questions Chapter 1 28 PDF book includes high school question papers to review practice tests for exams A Level Chemistry Practice Tests a textbook s revision guide with chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam A Level Chemistry Questions Bank Chapter 1 28 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Alcohols and Esters Questions Chapter 2 Atomic Structure and Theory Questions Chapter 3 Benzene Chemical Compound Questions Chapter 4 Carbonyl Compounds Questions Chapter 5 Carboxylic Acids and Acyl Compounds Questions Chapter 6 Chemical Bonding Questions Chapter 7 Chemistry of Life Questions Chapter 8 Electrode Potential Questions Chapter 9 Electrons in Atoms Questions

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