

## CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

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### Multiple Choice

*Identify the choice that best completes the statement or answers the question.*

- \_\_\_\_\_ 1) Which of the following observations is quantitative?  
a. The liquid turns blue litmus paper red.  
b. The liquid boils at 100°C.  
c. The liquid tastes bitter.  
d. The liquid is cloudy.
- \_\_\_\_\_ 2) The symbols for units of length in order from largest to smallest are  
a. m, cm, mm, km. c. km, mm, cm, m.  
b. mm, m, cm, km. d. km, m, cm, mm.
- \_\_\_\_\_ 3) A volume of 1 milliliter is equivalent to  
a. 1 cubic centimeter. c. 1 liter.  
b. 1 gram. d.  $10^{-3}$  cubic decimeters.
- \_\_\_\_\_ 4) Which pair of quantities determines the density of a material?  
a. mass and weight c. volume and concentration  
b. volume and weight d. volume and mass
- \_\_\_\_\_ 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm<sup>3</sup>. The density of gold is  
a. 0.0518 g/cm<sup>3</sup>. c. 101.5 g/cm<sup>3</sup>.  
b. 19.3 g/cm<sup>3</sup>. d. 483 g/cm<sup>3</sup>.
- \_\_\_\_\_ 6) The number of grams equal to 0.5 kg is  
a. 0.0005. c. 500.  
b. 0.005. d. 5000.
- \_\_\_\_\_ 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?  
a. 0.0706 cm c. 30.5 cm  
b. 14.2 cm d. 91.4 cm
- \_\_\_\_\_ 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have  
a. good precision.  
b. good accuracy.  
c. poor precision.  
d. poor accuracy.
- \_\_\_\_\_ 9) The measurement 0.035550 g rounded off to two significant figures would be  
a. 0.03 g. c. 0.036 g.  
b. 0.35 g. d.  $3.5 \times 10^2$  g.
- \_\_\_\_\_ 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

# Chemistry Final Exam Study Guide Problems

**Sheila Tobias, Jacqueline Raphael**



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## **Table of Contents Chemistry Final Exam Study Guide Problems**

1. Understanding the eBook Chemistry Final Exam Study Guide Problems
  - The Rise of Digital Reading Chemistry Final Exam Study Guide Problems
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Final Exam Study Guide Problems
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Final Exam Study Guide Problems
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Final Exam Study Guide Problems
  - Personalized Recommendations
  - Chemistry Final Exam Study Guide Problems User Reviews and Ratings
  - Chemistry Final Exam Study Guide Problems and Bestseller Lists
5. Accessing Chemistry Final Exam Study Guide Problems Free and Paid eBooks
  - Chemistry Final Exam Study Guide Problems Public Domain eBooks
  - Chemistry Final Exam Study Guide Problems eBook Subscription Services
  - Chemistry Final Exam Study Guide Problems Budget-Friendly Options
6. Navigating Chemistry Final Exam Study Guide Problems eBook Formats

- ePub, PDF, MOBI, and More
- Chemistry Final Exam Study Guide Problems Compatibility with Devices
- Chemistry Final Exam Study Guide Problems Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Final Exam Study Guide Problems
  - Highlighting and Note-Taking Chemistry Final Exam Study Guide Problems
  - Interactive Elements Chemistry Final Exam Study Guide Problems
- 8. Staying Engaged with Chemistry Final Exam Study Guide Problems
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Final Exam Study Guide Problems
- 9. Balancing eBooks and Physical Books Chemistry Final Exam Study Guide Problems
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Final Exam Study Guide Problems
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Final Exam Study Guide Problems
  - Setting Reading Goals Chemistry Final Exam Study Guide Problems
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Final Exam Study Guide Problems
  - Fact-Checking eBook Content of Chemistry Final Exam Study Guide Problems
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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